



Crystallographic methods: Historical perspectives

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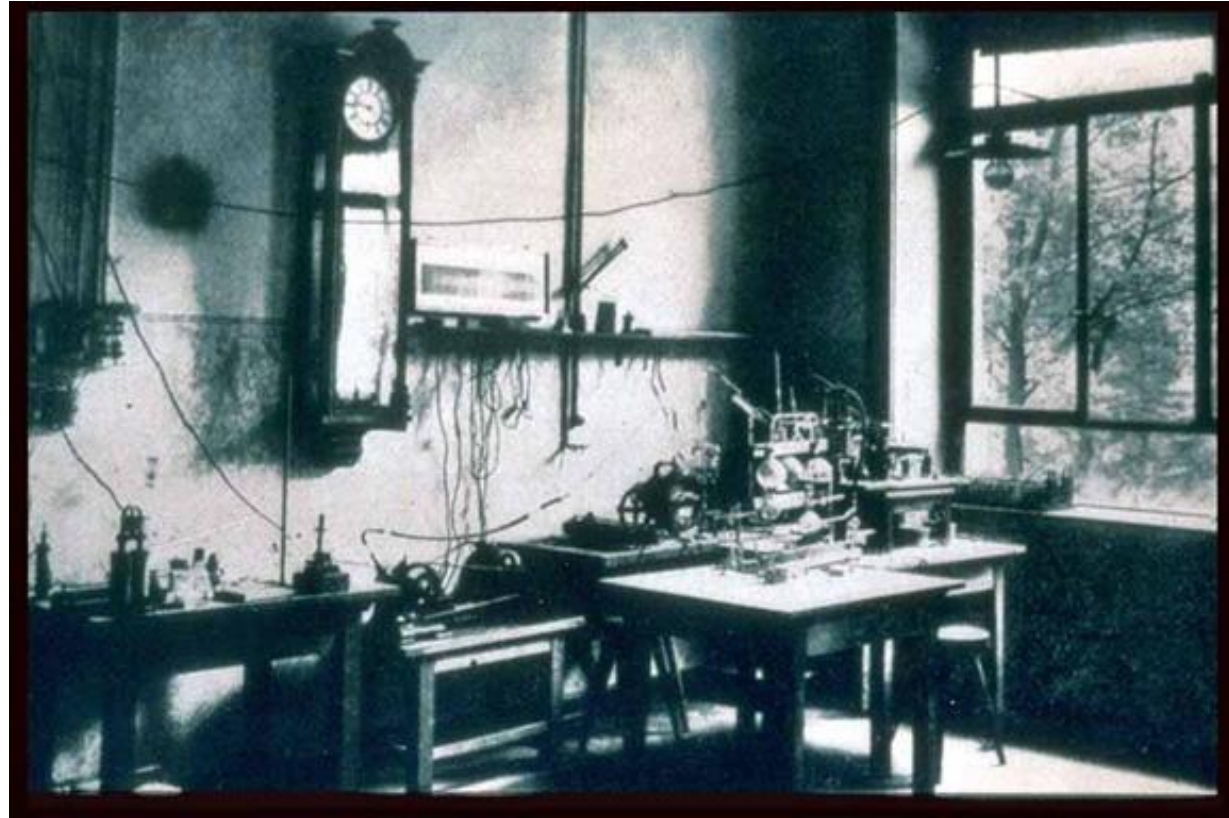
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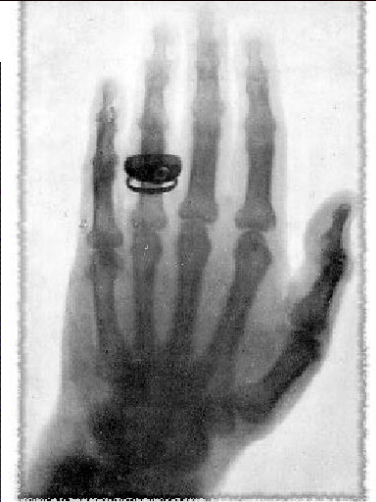
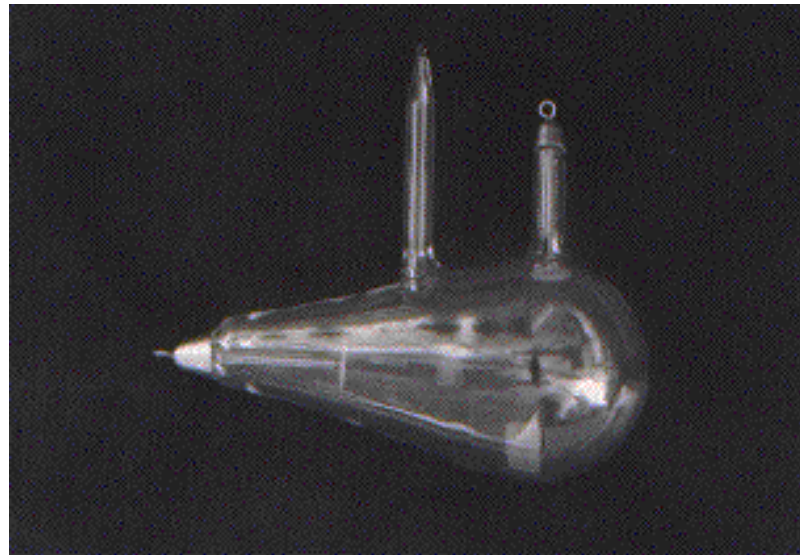
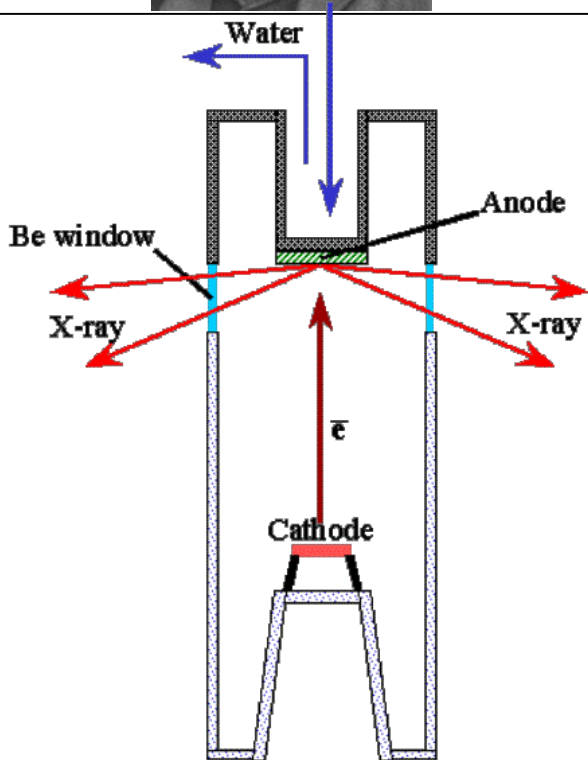
15 August 2017

Happy independence day!

History began in 1895 with Wilhelm Conrad Röntgen



Crooke's tube



First X ray made in public. Hand of the famed anatomist, Albert von Kölliker, made during Roentgen's initial lecture before the Würzburg Physical Medical Society on January 23, 1896.

Wave or particle?

X-ray : waves or particules

Theoretical evidence of radiation emission related to the slowing down of charged particles (J.J. Thomson , 1896)

→ « Theoretical discovery of X-rays »



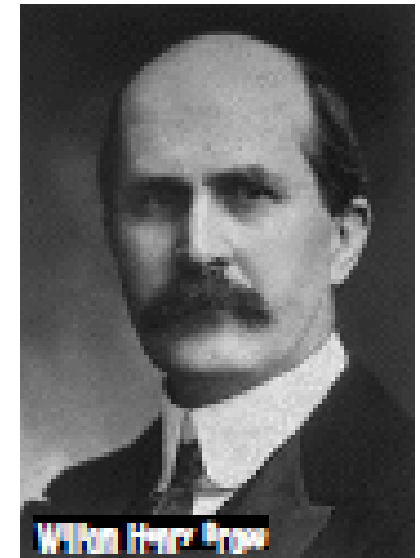
Joseph John Thomson

Physic Nobel price, 1906

P. Lenard, B. Walter, J.J.Thomson

waves or particles ?

W.H. Bragg



William Henry Bragg

Physic Nobel price, 1915

Particles or waves?

W.L Bragg and W.H. Bragg

When a wave hits an object, the region immediately behind that object is protected from the wave (shadow). Shadows are sharper close to an object than they are further from it. The wave that just missed the object spreads in a circle or sphere, into the space behind the object. This is why shadows become more blurred further away from the object that casts them.

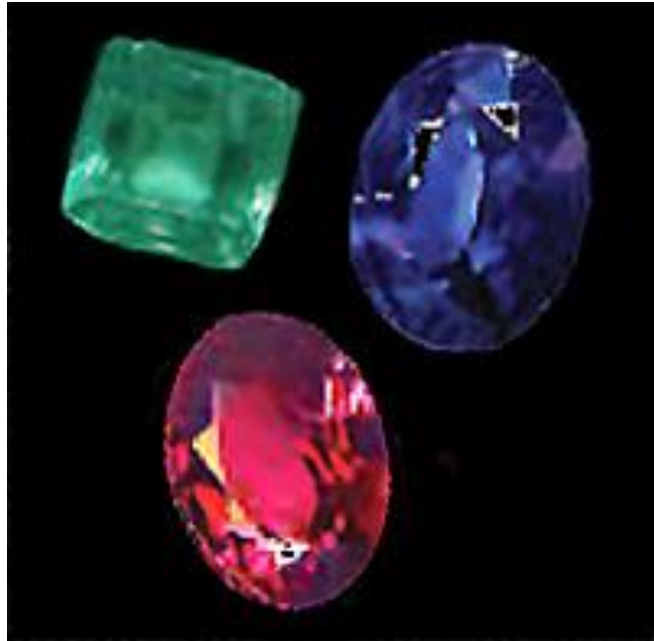
Wilhelm Wien

wavelength of X-rays should be around one hundredth of a nanometre

Lovely quartz crystals in Nature's kitchen



Crystals: symbols of beauty as well as vanity



In ancient India, crystallography was called

“Ratnashastra”

Origin of gems

भूस्वभावाद्धि रत्नानि जातानि विविधानि तु ।
उपला रत्नरूपत्वं प्राप्ताः कालान्तरेण वा ॥

Due to earth's nature, various forms of gems
are formed over long incubation periods

पुरा पृथिव्या रत्नानि गर्भे त्वासन्धि सर्वशः ।
रत्नगर्भा इति सा भूमिः ख्याताभूद् भुवनत्रये ॥

Because of the presence of gems in its bosom
Earth has earned the name “RATNAGARBHA”

Classification of gems

Maharatna

(precious stones)

Vajra (diamond)

Mukta (pearl)

Manikya (Ruby)

Indranila (Sapphire)

Uparatna

(semi precious stones)

Marakata (Emerald)

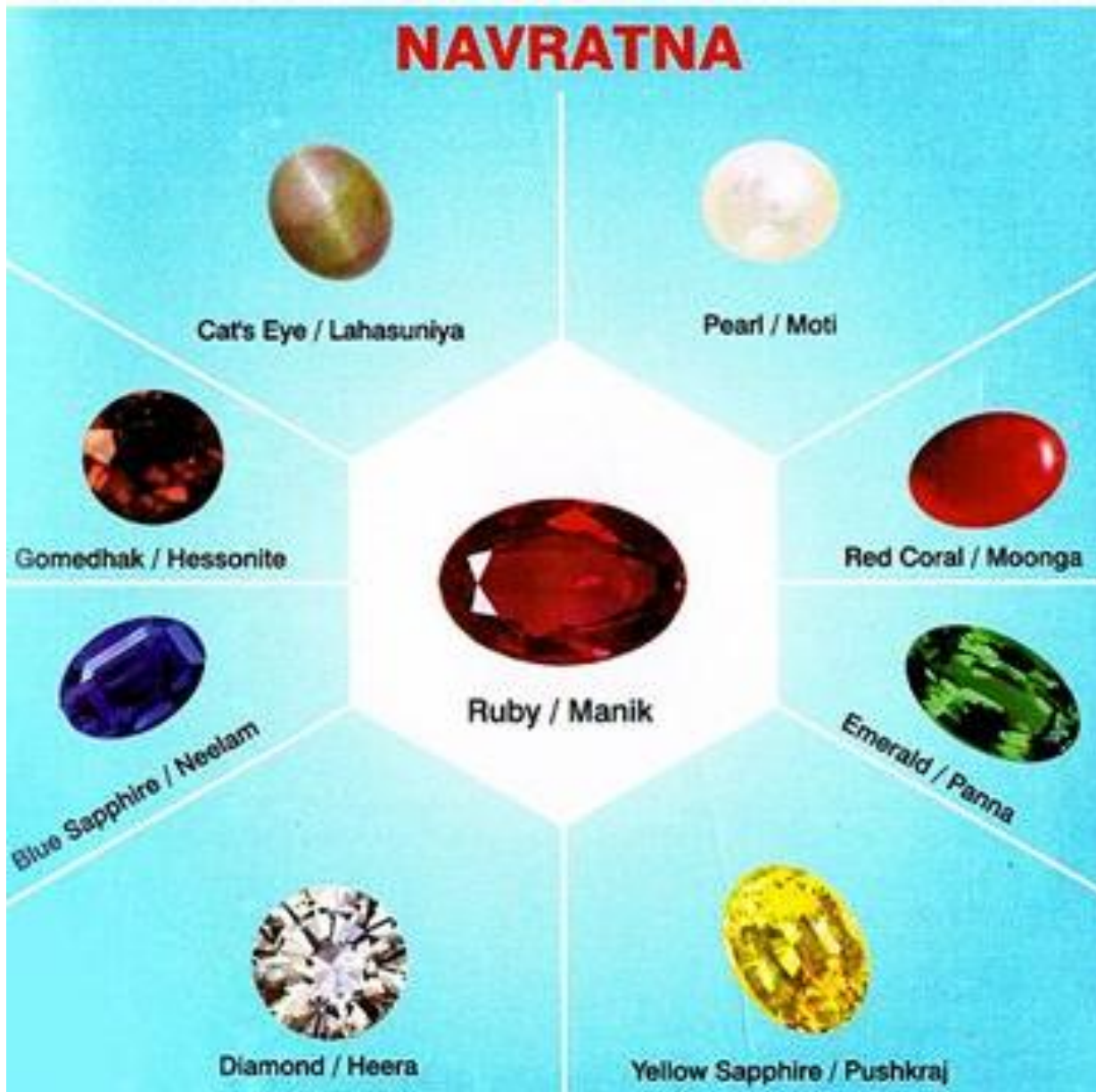
Vidruma (Coral)

Gomeda (Jacinth)

Pushparaga (Topaz)

Vaidurya (Lapis Lazuli)

One possessed of all desirable qualities



VAJRA (Diamond)

It was well recognized that diamond
is the hardest substance

षट्कोणत्वं लघुत्वं च समाष्टदलता तथा
तीक्ष्णाग्रहा निर्मलत्वं इति पञ्च गुणाः स्मृताः ॥

Diamond has six vertices and eight facets

Octahedral symmetry

Five properties of diamond

Vajrakaya

VARAHAMIHIRA studied crystal defects

Diamond is the hardest gems

पृथिव्यां यानि रत्नानि ये चान्ये लोहधातवः ।
सर्वं ताद्विलिखेत्परं वपुः तैर्न वलिख्यते ॥

Halasya Mahatyam

**Of all the gems on earth, and
all materials made of metals,
the hardest is the diamond**

Diamond cannot be scratched by any other object

Experimental examination of gems

Diamond

Pounding

Emerald

Green ray scattering when crystal faces sun

Sapphire

Blue when placed in milk

Imagination unlimited

Sun

Ruby

Venus

Diamond

Mars

Coral

Jupiter

Yellow Sapphire

Saturn

Blue Sapphire

Mercury

Emerald

Moon

Pearl

Rahu

Hessonite Garnet

Ketu

Cat's Eye Chrysoberyl

Crystals : periodic arrangement of molecules



Magic numbers of crystallography

7 crystal systems

14 Bravais lattices

11 Laue groups

32 Point groups

17 Plane groups

230 Space groups

65 Non-centric space groups

The mathematical principles that underlie these numbers were understood before the birth of X-ray crystallography

Laue, great synthesis of ideas



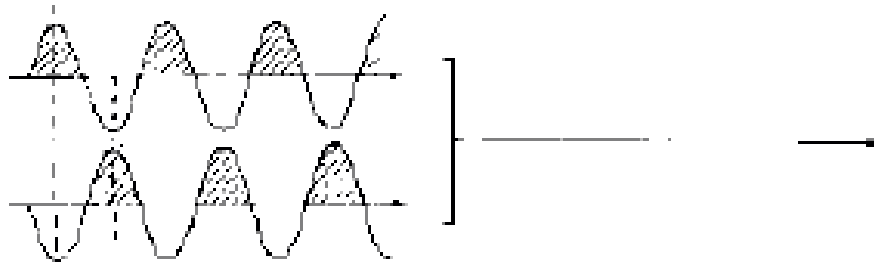
Physic Nobel price, 1914



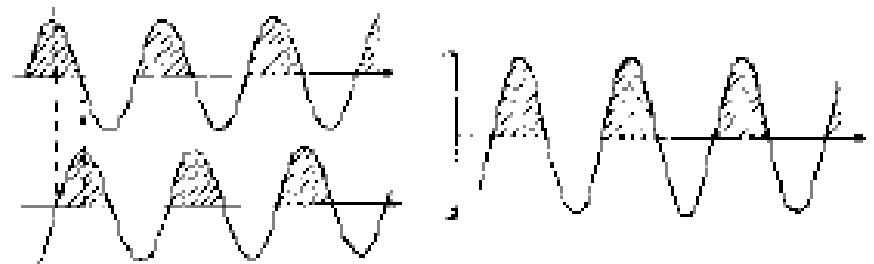
What matters in adding two waves is the phase difference



**In phase
constructive**

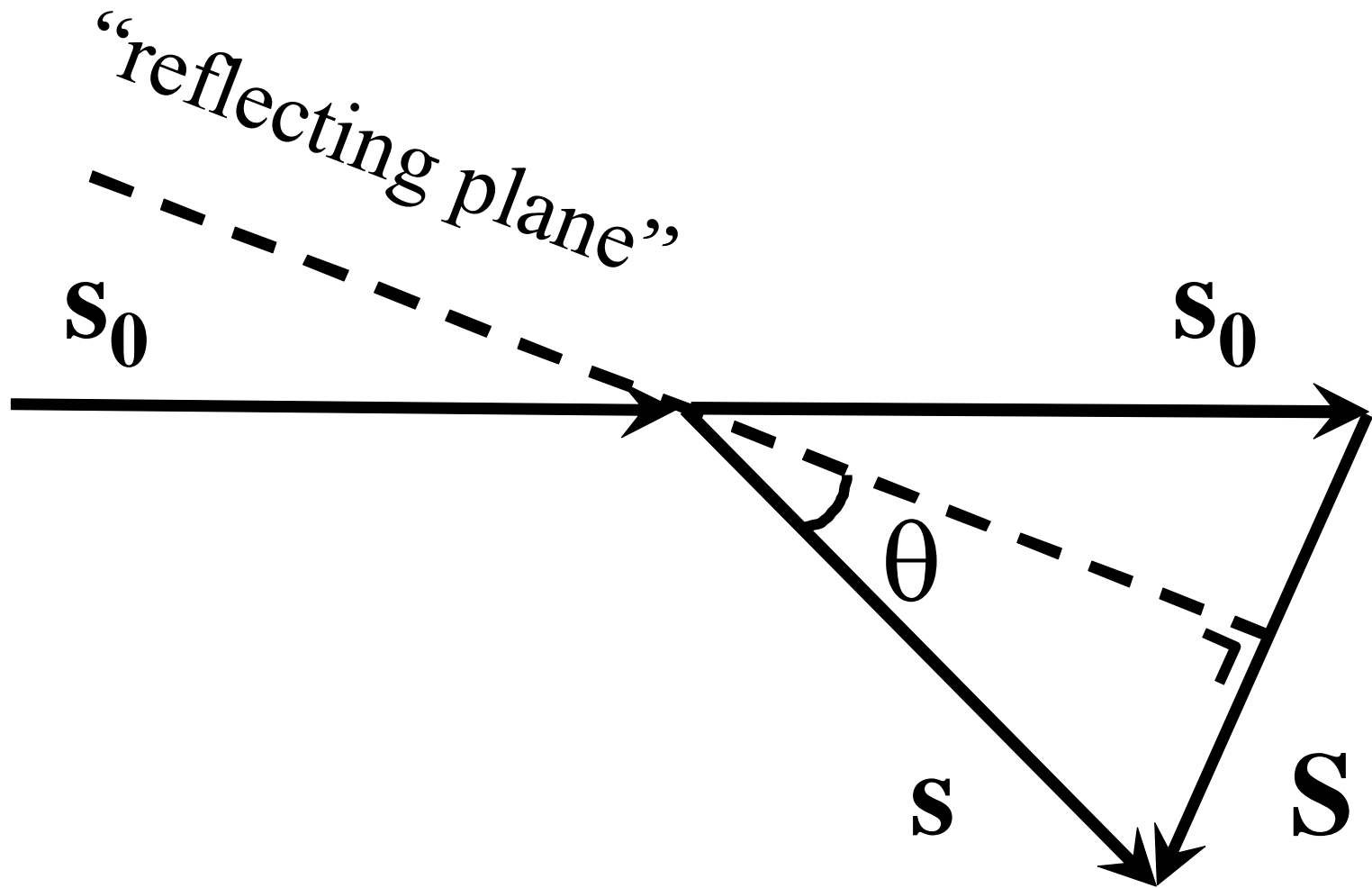


**Out of Phase
Destructive**



**Partially out of phase
Intermediate amplitude**

Scattering vector



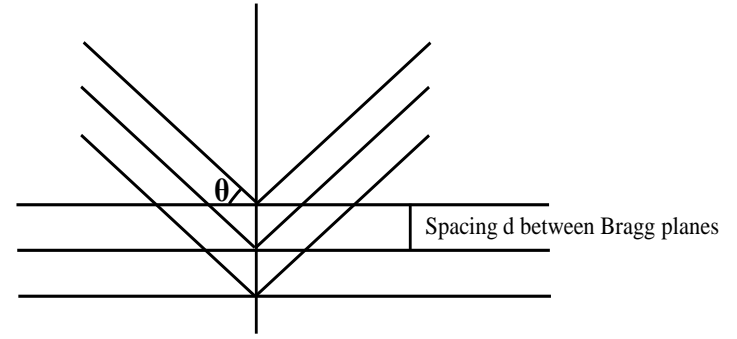
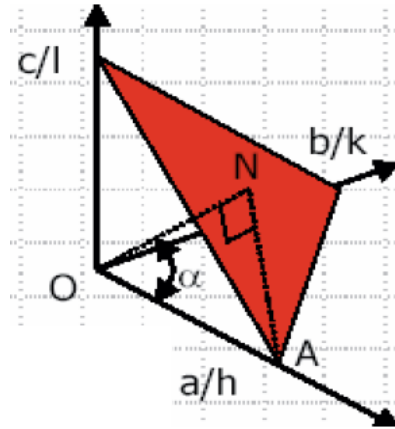
Three ways of describing interaction of crystals with X-rays

$$\mathbf{a} \cdot \mathbf{S} = h$$

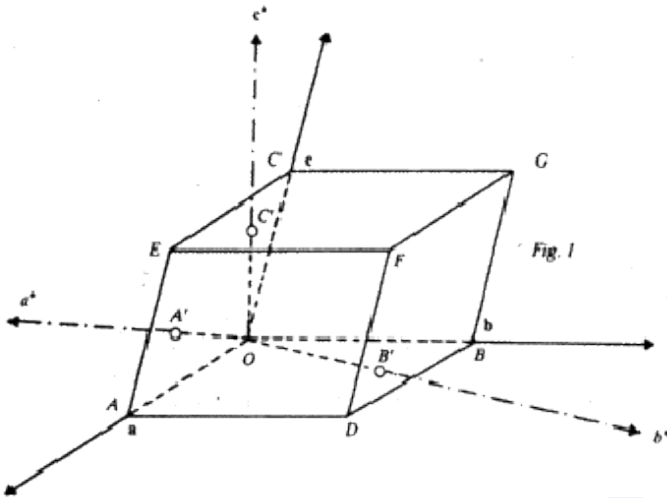
$$\mathbf{b} \cdot \mathbf{S} = k$$

$$\mathbf{c} \cdot \mathbf{S} = l$$

Laue conditions



$$\text{Bragg's law: } 2 d \sin \theta = \lambda$$

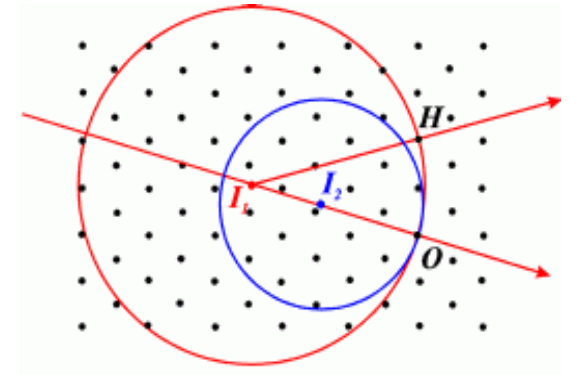


$$\mathbf{a}^* = \mathbf{b} \times \mathbf{c} / (\mathbf{a} \cdot \mathbf{b} \times \mathbf{c})$$

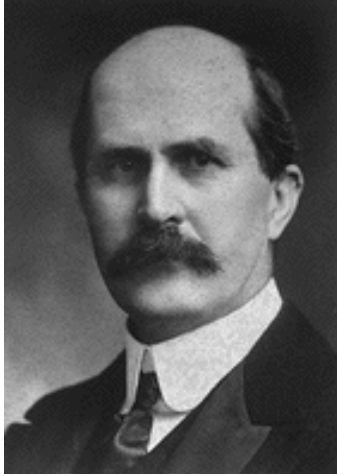
$$\mathbf{b}^* = \mathbf{c} \times \mathbf{a} / (\mathbf{a} \cdot \mathbf{b} \times \mathbf{c})$$

$$\mathbf{c}^* = \mathbf{a} \times \mathbf{b} / (\mathbf{a} \cdot \mathbf{b} \times \mathbf{c})$$

Ewald's construction

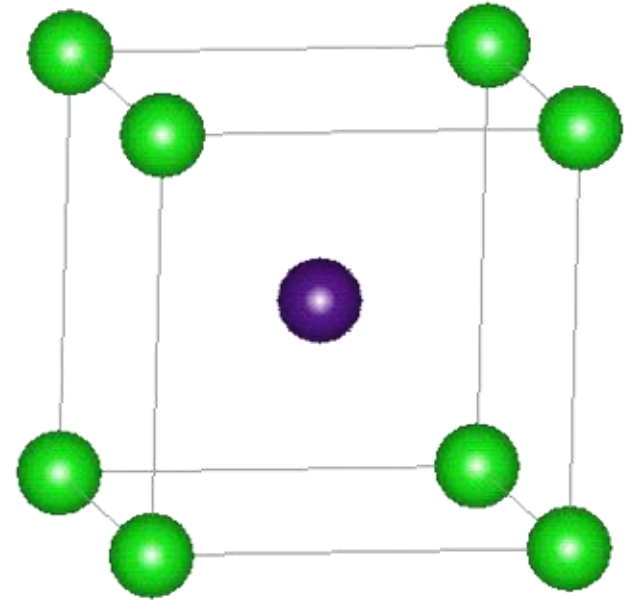
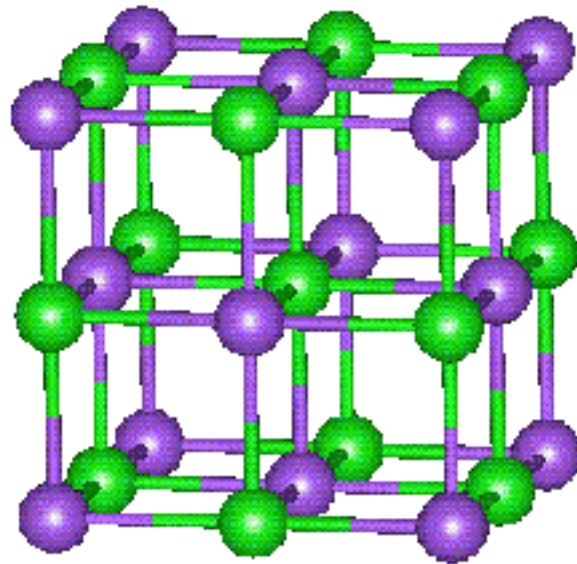


**Sir William
Henry Bragg
(1862-1942)**



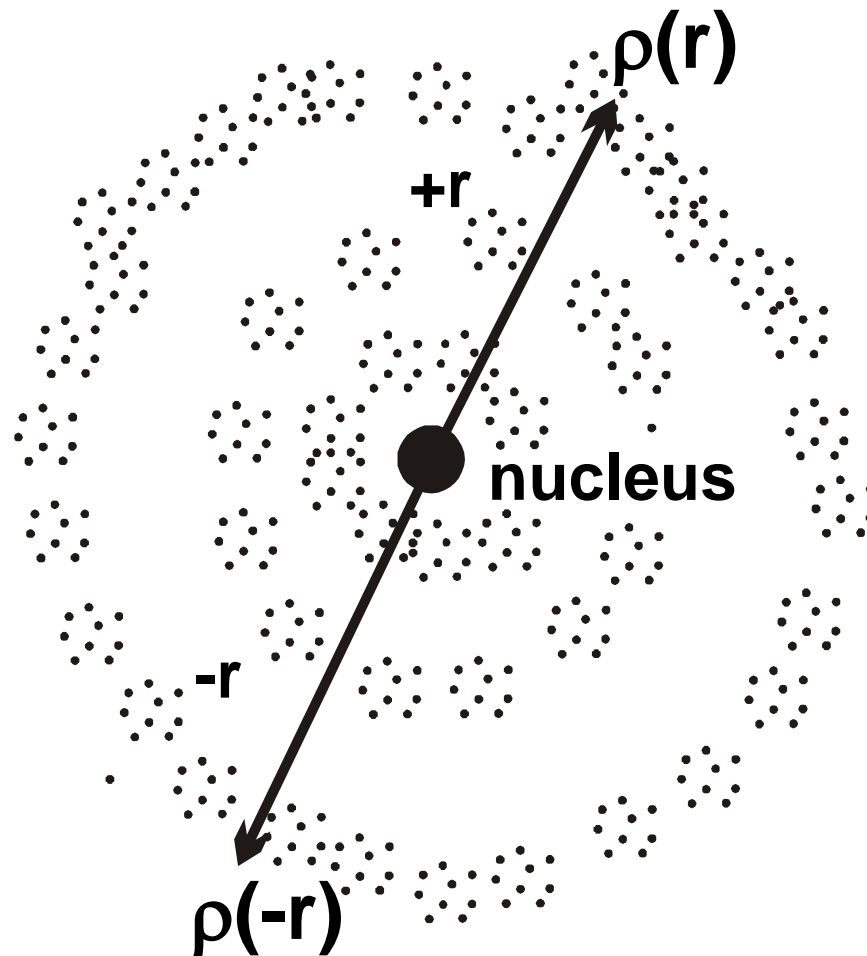
**Sir William
Lawrence Bragg
(1890-1971)**

A humble beginning

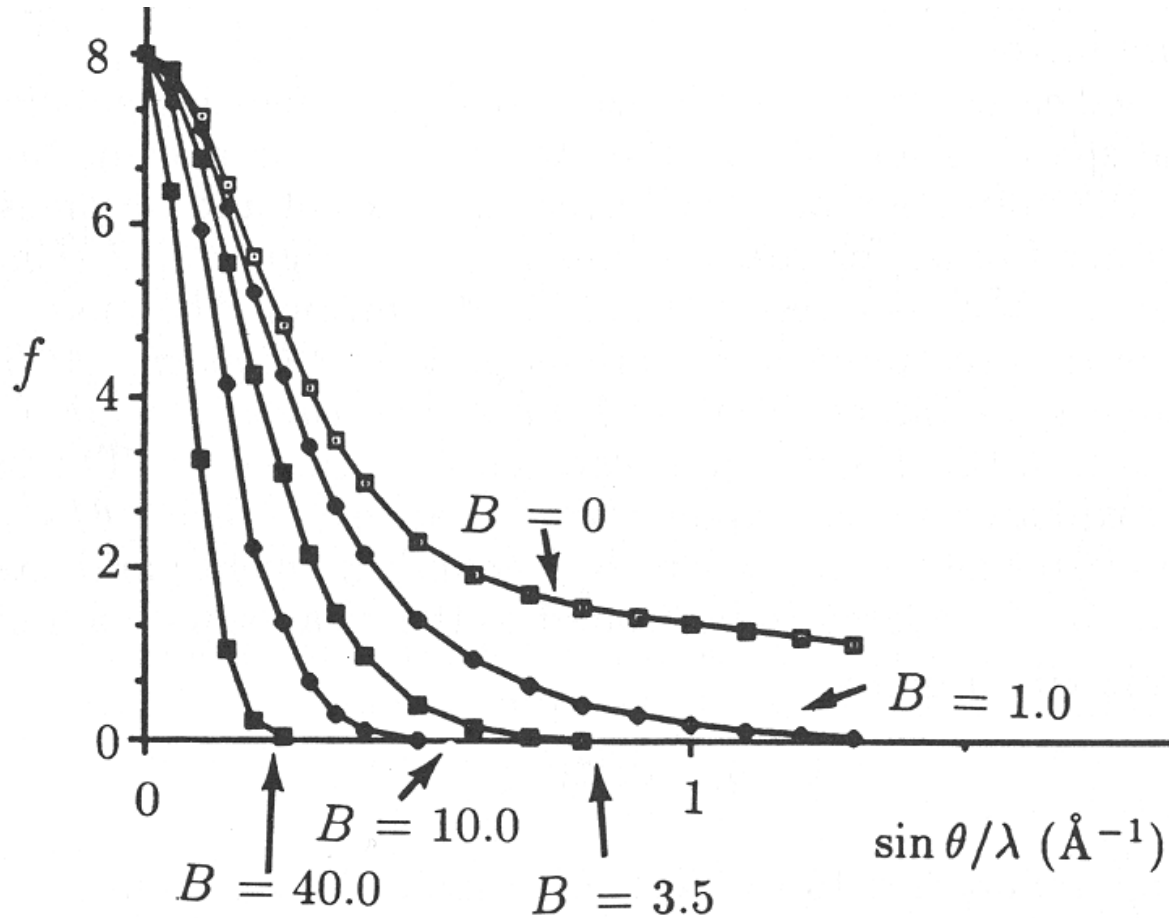


Structure of sodium chloride

Scattering by an Atom



Atomic scattering factor



$$f = f_0 \exp(-B \sin^2 \theta / \lambda) : \text{phase factor}$$

Scattering by a Unit Cell

$$\mathbf{f}_j = f_j e^{2\pi i \mathbf{r}_j \cdot \mathbf{S}}$$

$$\mathbf{F}(\mathbf{S}) = \sum_{j=1}^n f_j e^{2\pi i \mathbf{r}_j \cdot \mathbf{S}}$$

This expression assumes X-ray beam scattered by a fictitious atom placed at the origin of the unit cell as the reference beam

**X-ray diffraction yields the unit cell transform
sampled at integral reciprocal lattice points**



Fundamental Equations of Crystallography

Structure Factor

$$F(\mathbf{hkl}) = \iiint \rho(\mathbf{xyz}) e^{2\pi i(\mathbf{hx} + \mathbf{ky} + \mathbf{lz})} d\mathbf{x} d\mathbf{y} d\mathbf{z}$$

Electron density

$$\rho(\mathbf{xyz}) = \sum \sum \sum F(\mathbf{hkl}) e^{i\alpha} e^{-2\pi i(\mathbf{hx} + \mathbf{ky} + \mathbf{lz})}$$

History of first structures

Dorothy and Bernal took first pictures of protein crystals in the 30's

Max Perutz became a graduate student in the 20's

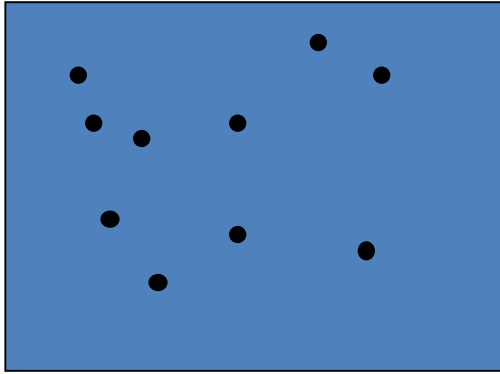
He demonstrated the way of determining phases in Green, Ingram and Perutz, JMB (1954)

Structure of myoglobin (1960), haemoglobin(1962)

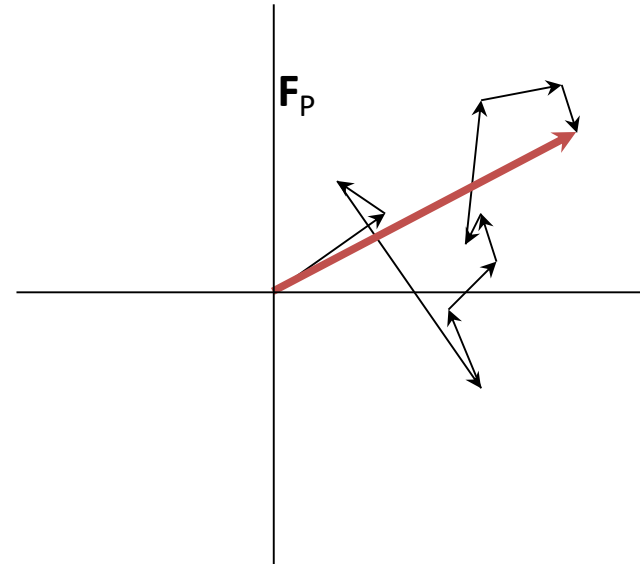
A true revolution

Today, we know more than 125,000 protein structures

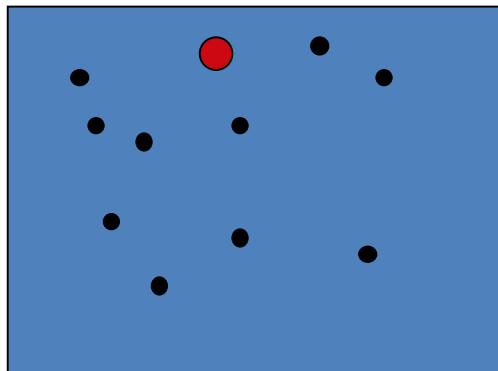
Crystal structure: Mild perturbation



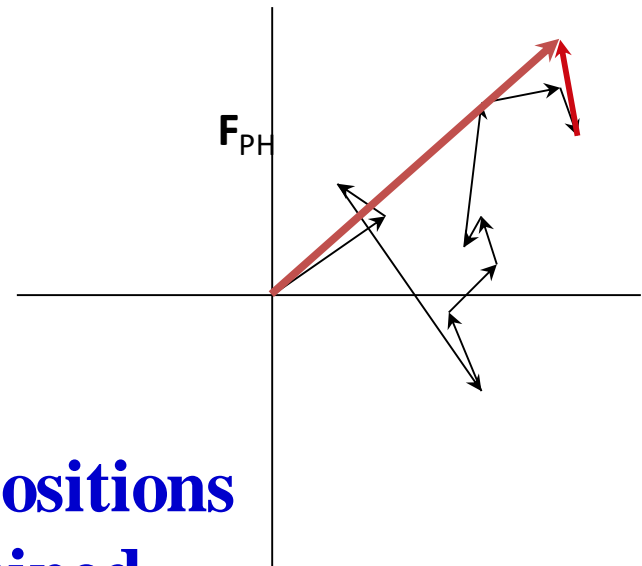
Native



Adding one (or more) atoms in known positions changes the structure factor in a known way



Derivative



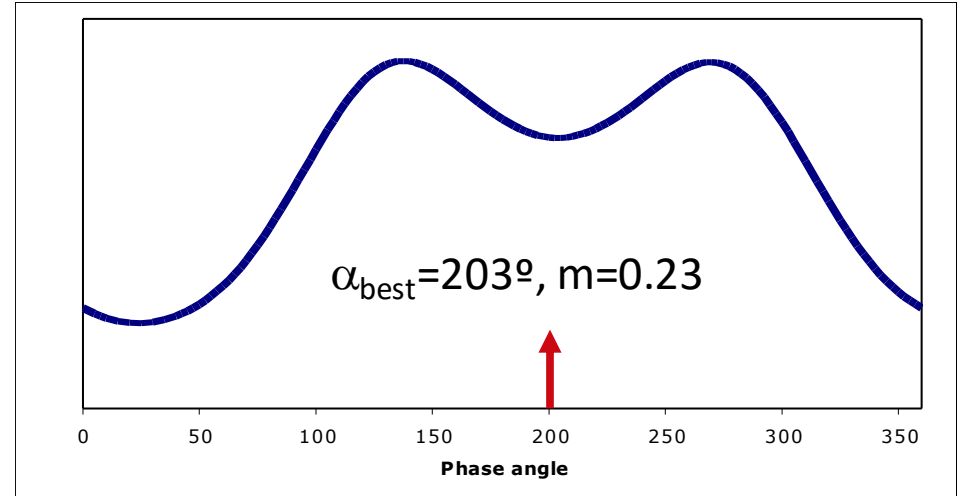
Heavy atom positions overdetermined

Resolving phase ambiguity

1 1 9 reflection

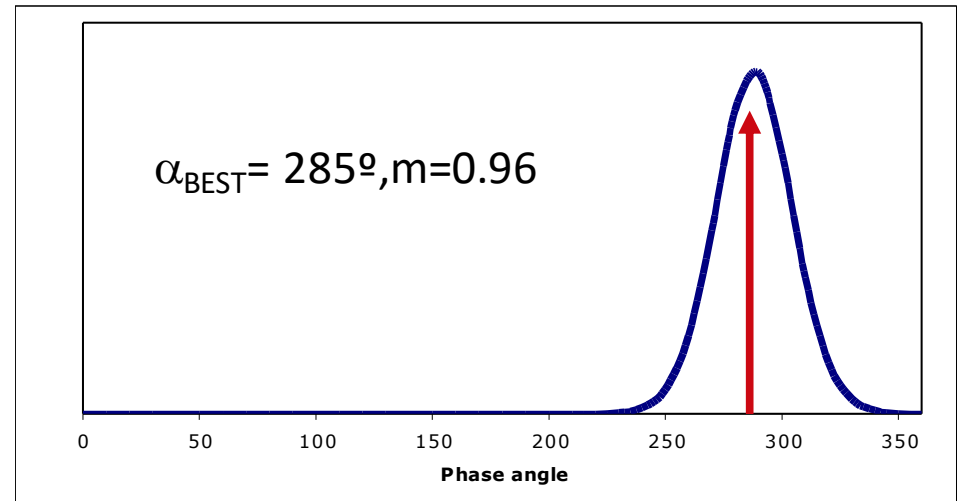
Blow and Crick

One derivative



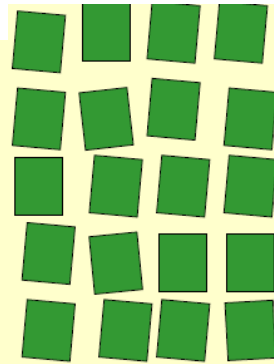
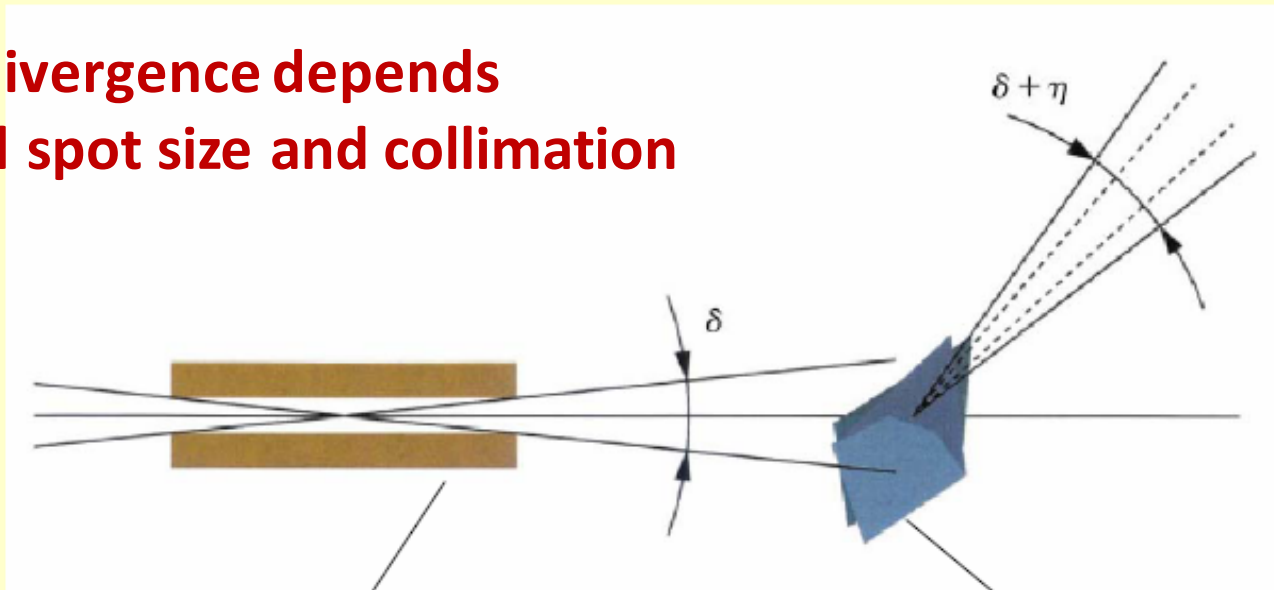
Three derivatives

$$P(\alpha_p) \propto \prod_{i=1} \exp(-\varepsilon_i^2 / 2E_i^2)$$



Crystal mosaicity and beam divergence determine the rocking curve

Beam divergence depends on focal spot size and collimation



Beam divergence

Crystal mosaicity

Alas! No control (usually)

Spot resolution problem: 1970's Home source

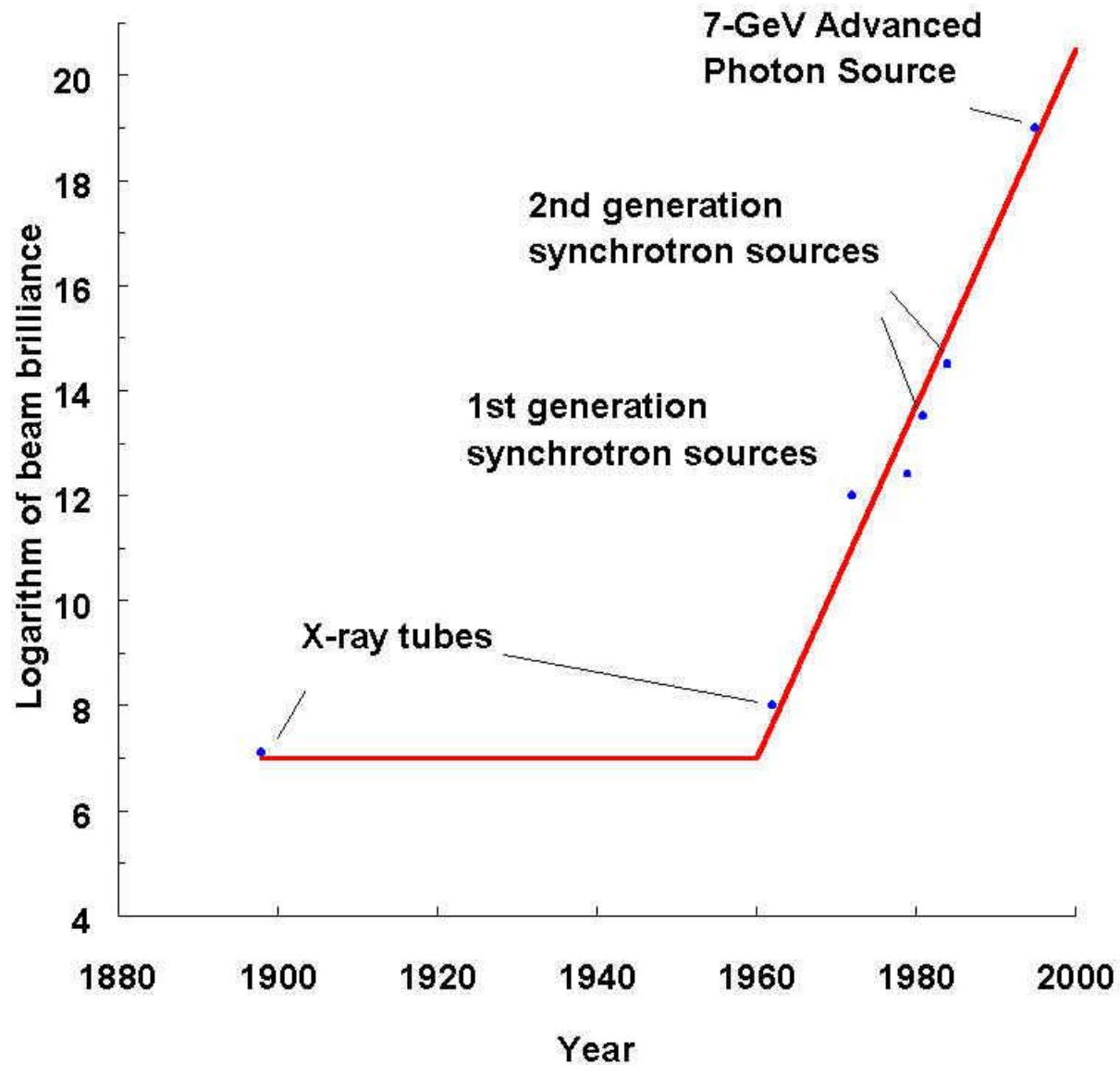


1980's synchrotrons





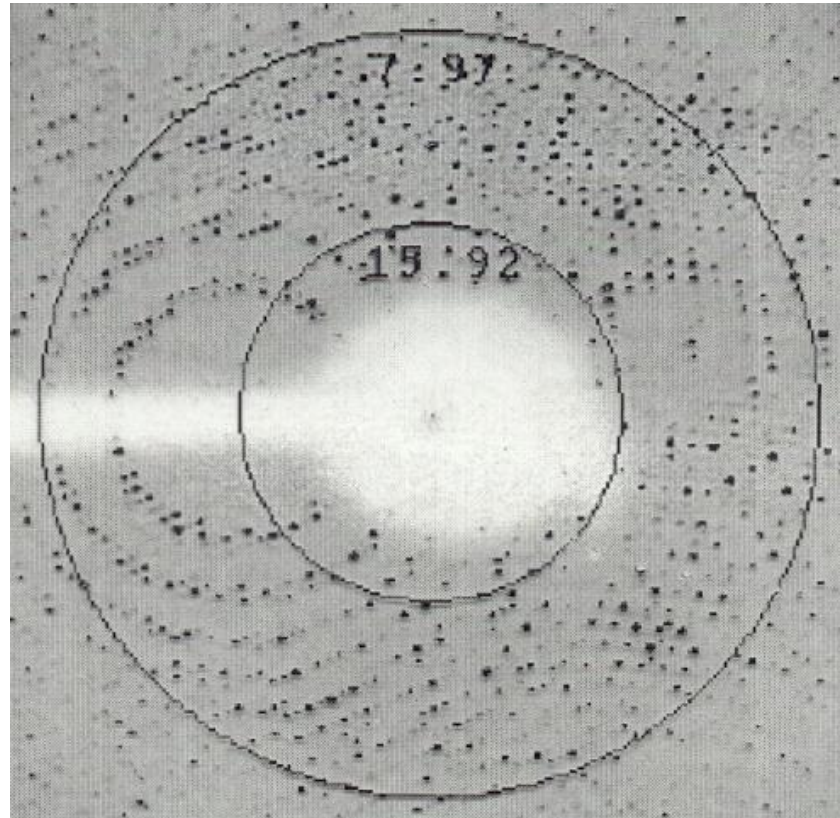
Unbelievable power



Recording intensities associated with three-dimensional reciprocal lattice on two-dimensional detectors

**Weisenberg
Precession**

Screenless oscillation photography



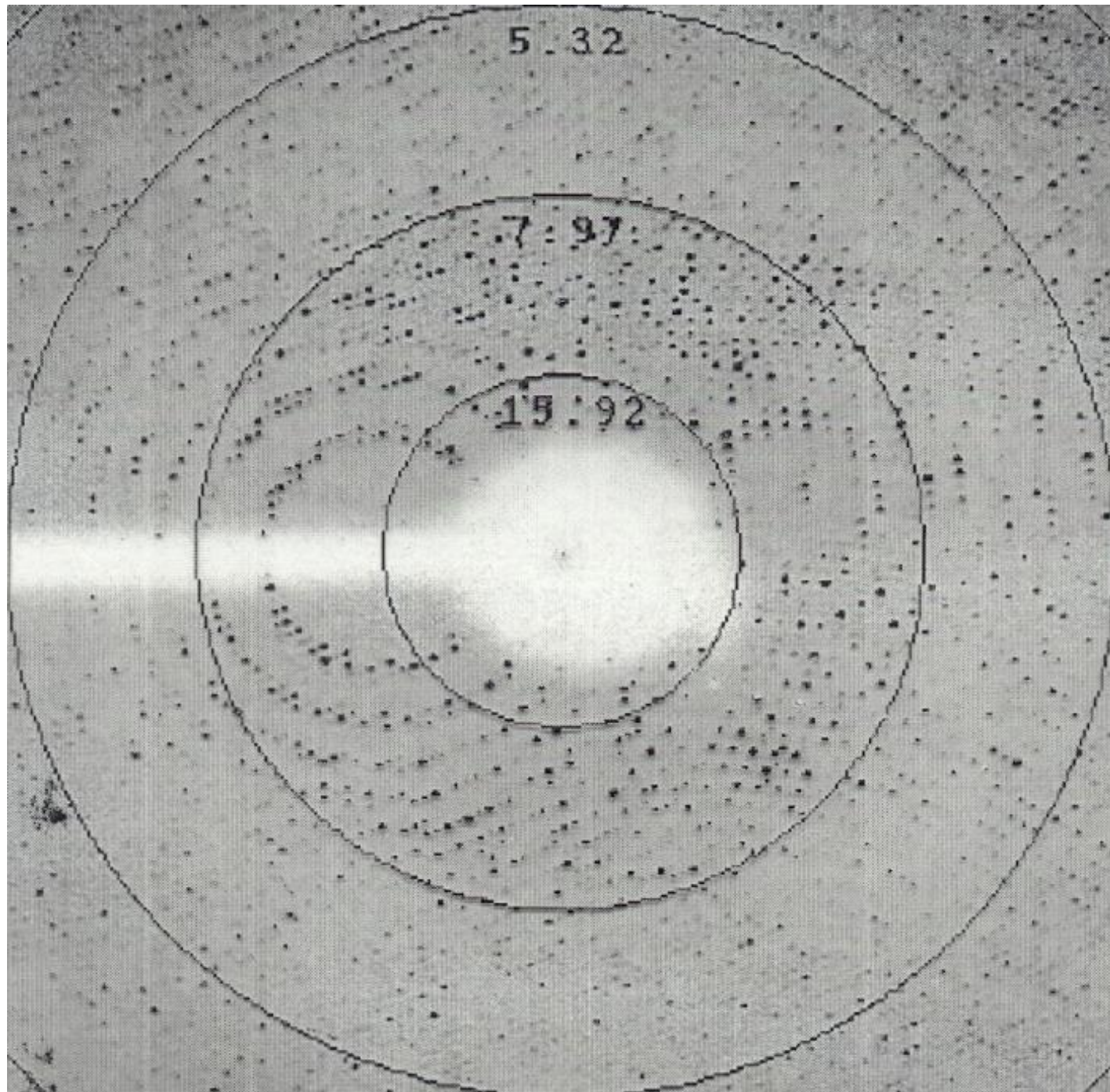


Twinkle twinkle
Little star
How I wonder
What you are

Where do you
come from?

How strong
are you?

How much do
I trust you?

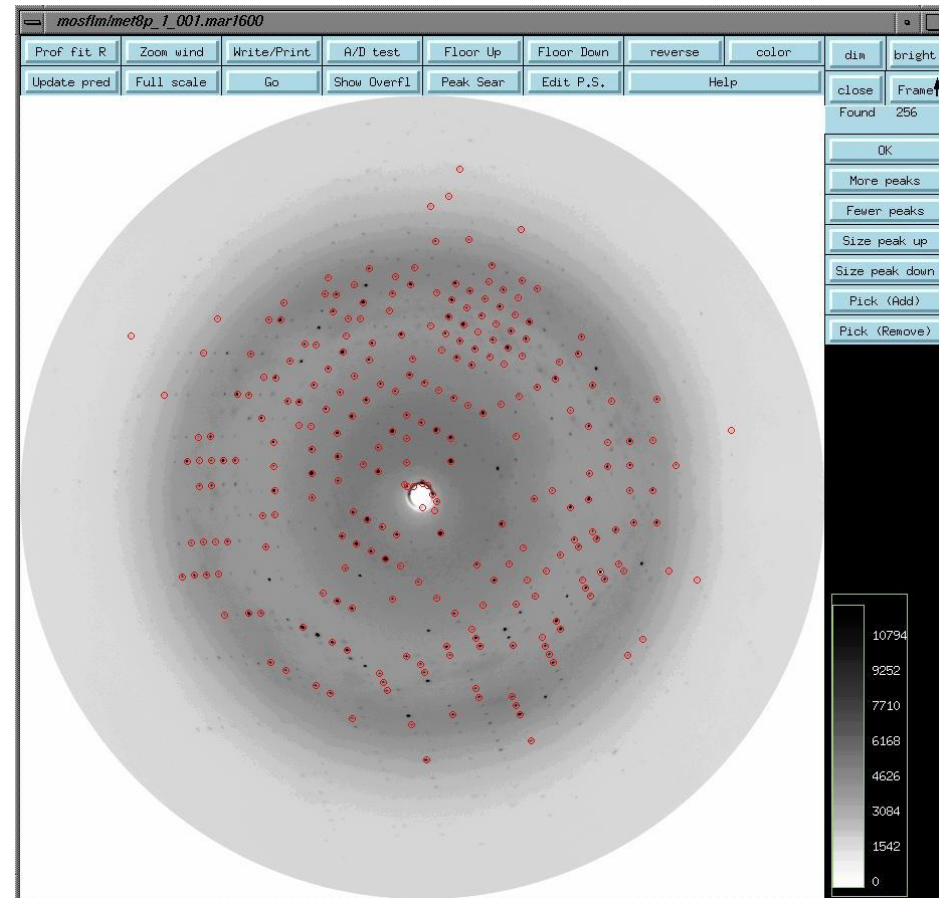


Auto-indexing

Difference vector method

**Projection of reciprocal
lattice points**

No need to pre-orient the crystal



Indexing

Refining orientation

Post-refinement

Bravais Lattice Table

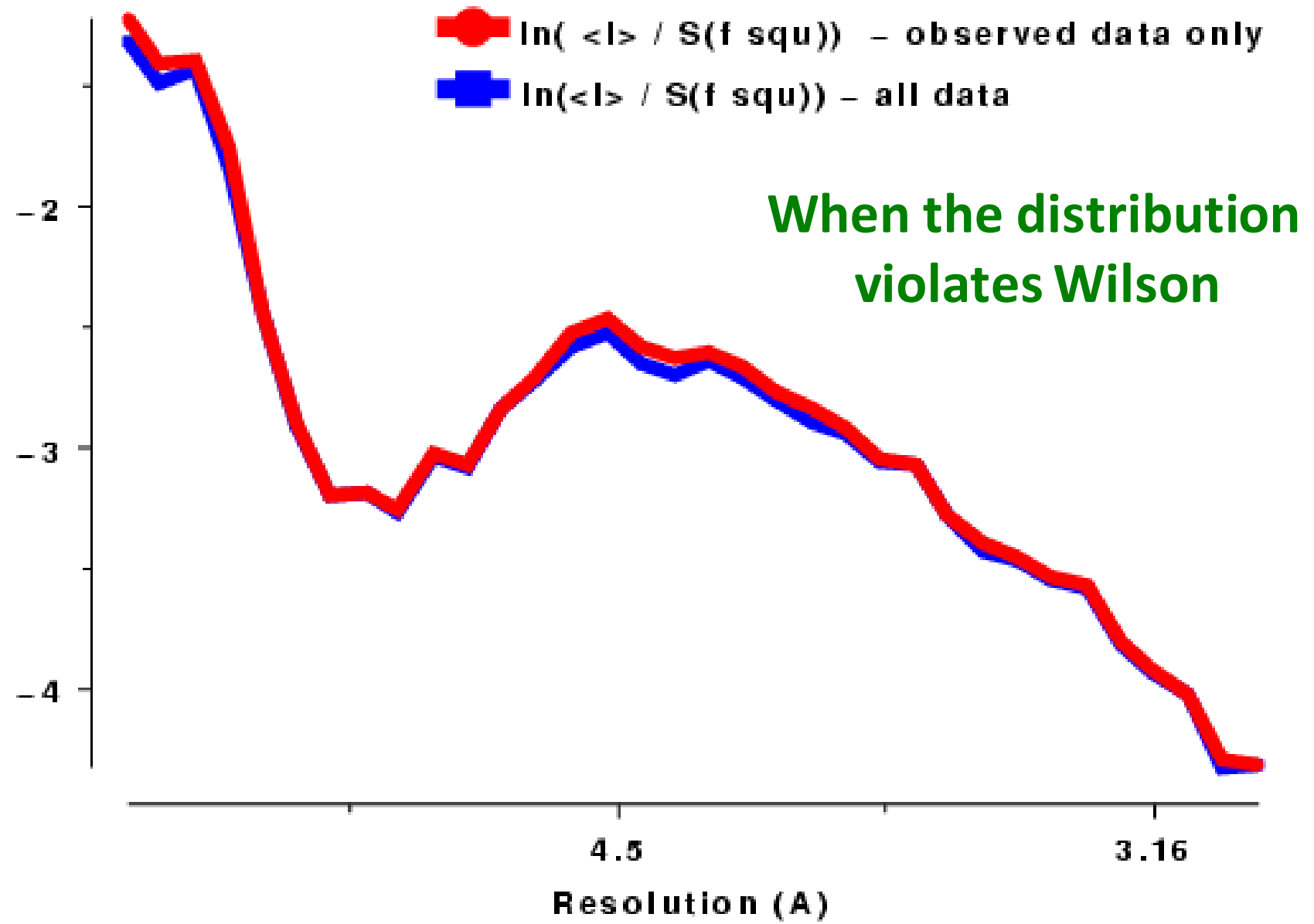
Autoindexing preformed for unit cell between 11.8 to 315 Angstroms

◇ primitive cubic	22.68%	45.48	90.68	90.77	60.16	90.06	90.05
		75.64	75.64	75.64	90.00	90.00	90.00
◇ I centred cubic	25.95%	101.67	90.77	101.41	63.61	78.46	116.65
		97.95	97.95	97.95	90.00	90.00	90.00
◇ F centred cubic	26.84%	101.67	101.69	163.42	97.07	97.18	126.85
		122.26	122.26	122.26	90.00	90.00	90.00
◇ primitive rhombohedral	7.82%	90.68	90.94	101.48	116.67	116.41	120.03
		94.37	94.37	94.37	117.70	117.70	117.70
		160.36	160.36	45.48	90.00	90.00	120.00
◆ primitive hexagonal	0.08%	90.68	90.94	45.48	90.02	90.05	120.03
		90.81	90.81	45.48	90.00	90.00	120.00
◇ primitive tetragonal	13.52%	90.68	90.77	45.48	89.94	90.05	119.84
		90.72	90.72	45.48	90.00	90.00	90.00
◇ I centred tetragonal	15.44%	163.42	90.94	45.48	89.98	73.90	90.06
		127.18	127.18	45.48	90.00	90.00	90.00
◇ primitive orthorhombic	13.52%	45.48	90.68	90.77	60.16	90.06	90.05
		45.48	90.68	90.77	90.00	90.00	90.00
◇ C centred orthorhombic	0.04%	90.94	157.02	45.48	89.94	89.98	89.93
		90.94	157.02	45.48	90.00	90.00	90.00
◇ I centred orthorhombic	6.79%	45.48	90.68	163.87	90.13	73.93	90.05
		45.48	90.68	163.87	90.00	90.00	90.00
◇ F centred orthorhombic	15.88%	45.48	186.94	187.10	58.19	75.99	75.96
		45.48	186.94	187.10	90.00	90.00	90.00
◇ primitive monoclinic	0.03%	90.68	45.48	90.77	90.06	119.84	89.95
		90.68	45.48	90.77	90.00	119.84	90.00
◇ C centred monoclinic	0.03%	157.02	90.94	45.48	89.98	90.06	90.07
		157.02	90.94	45.48	90.00	90.06	90.00
◇ primitive triclinic	0.00%	45.48	90.68	90.77	60.16	89.94	89.95

If you would like to change the crystal lattice: select desired bravais lattice, press Apply button and close window, otherwise just close window.

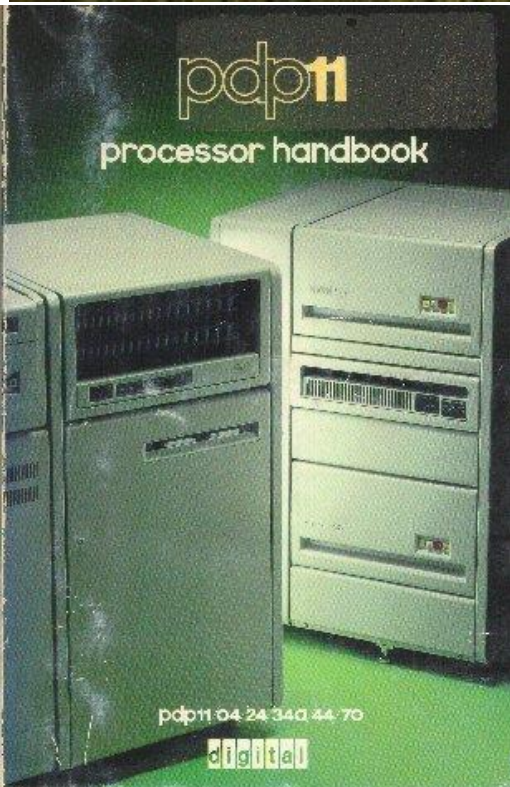
Apply Apply & Close

Wilson 1949: Statistics of intensities

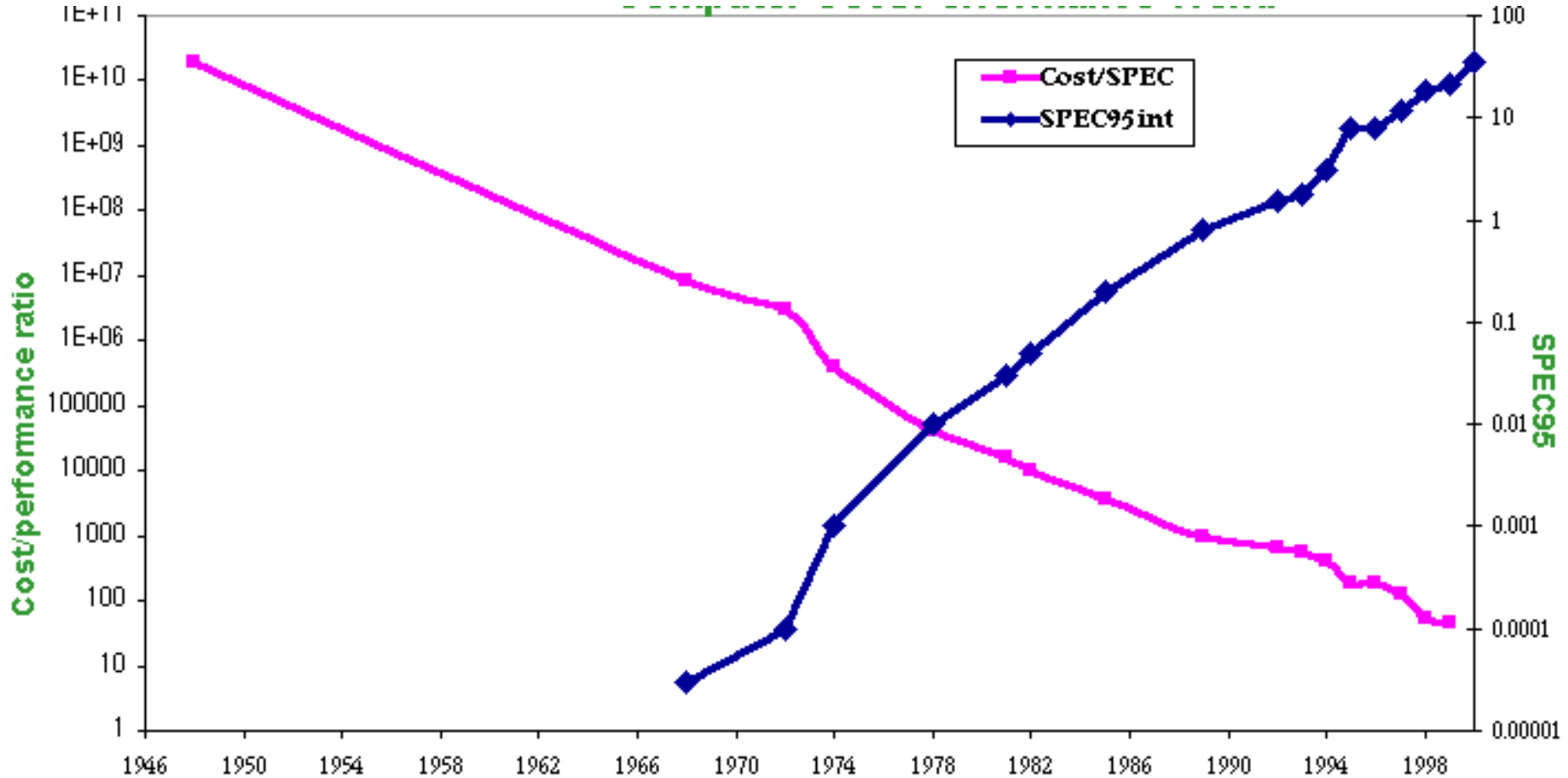




PDP 11/44: Dinosaurs size Walnut sized brain of DEC10



Computer cost and performance



Flash freezing: 1990

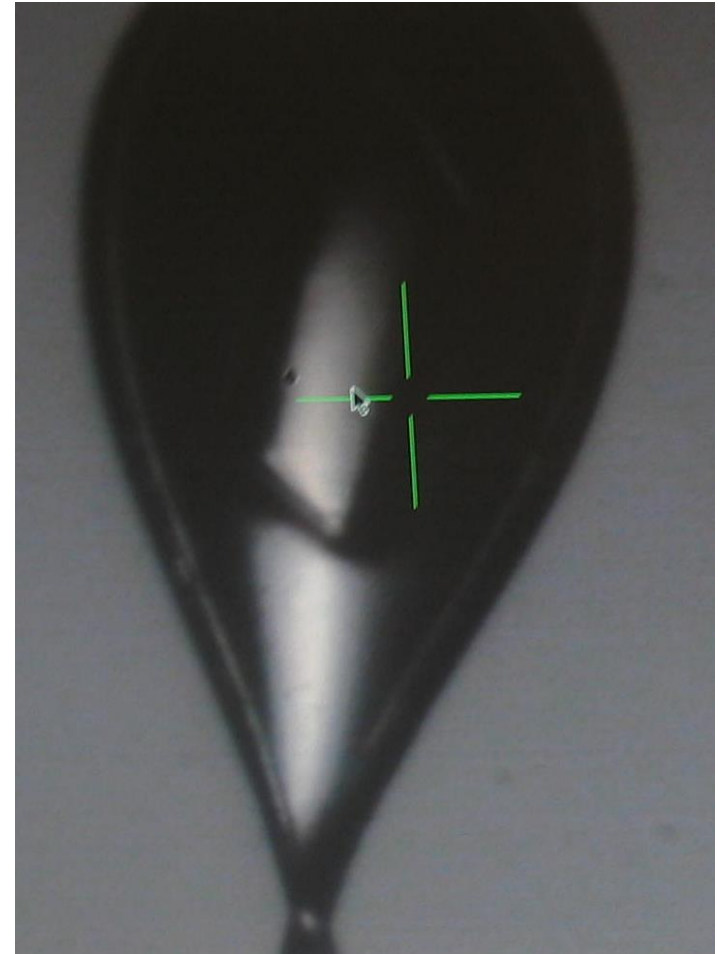
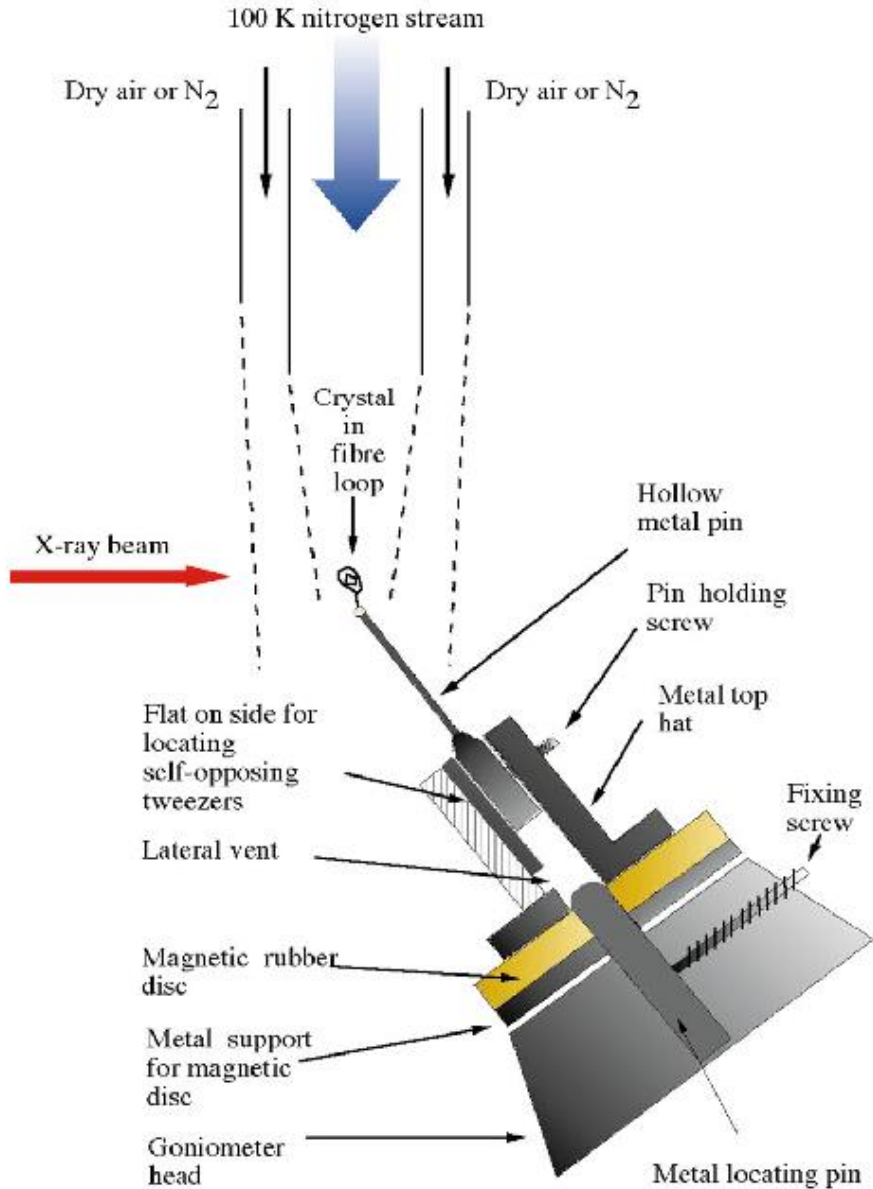
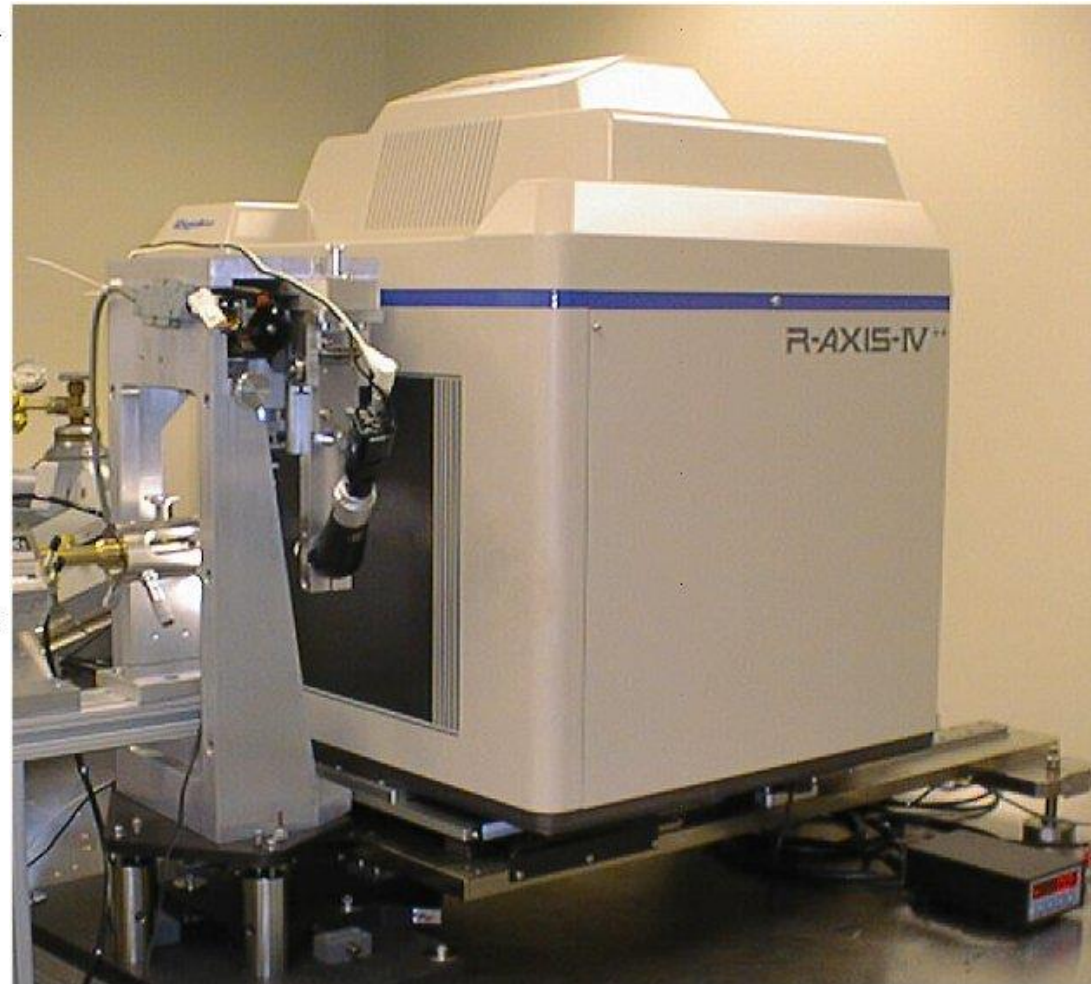
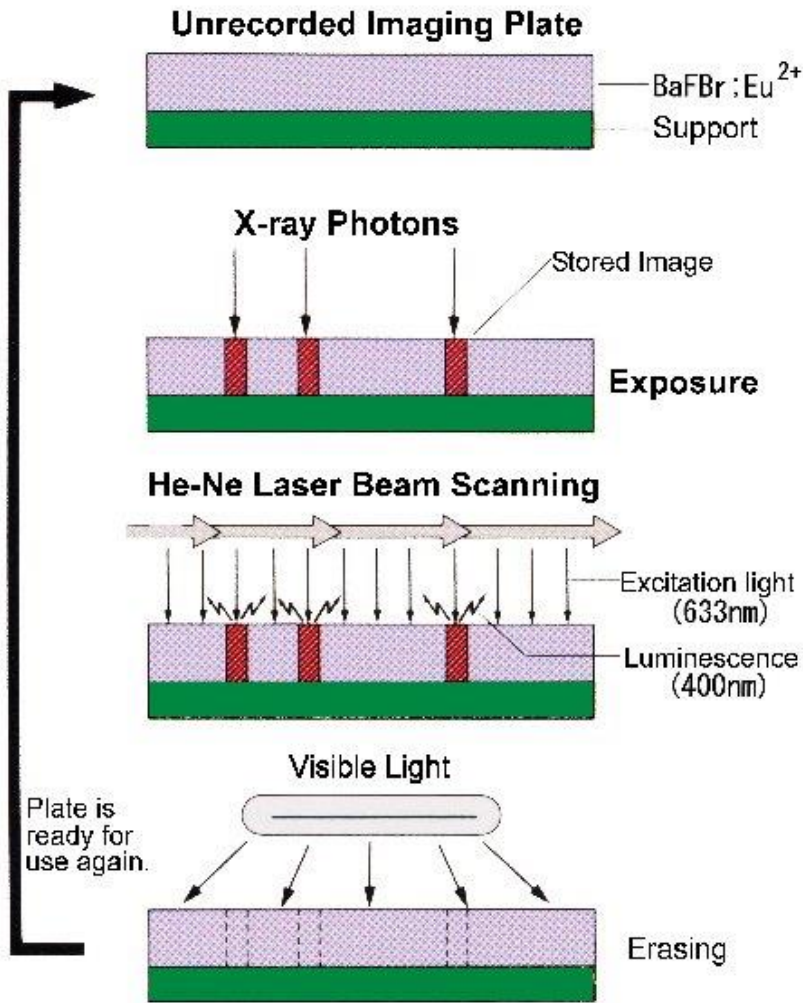
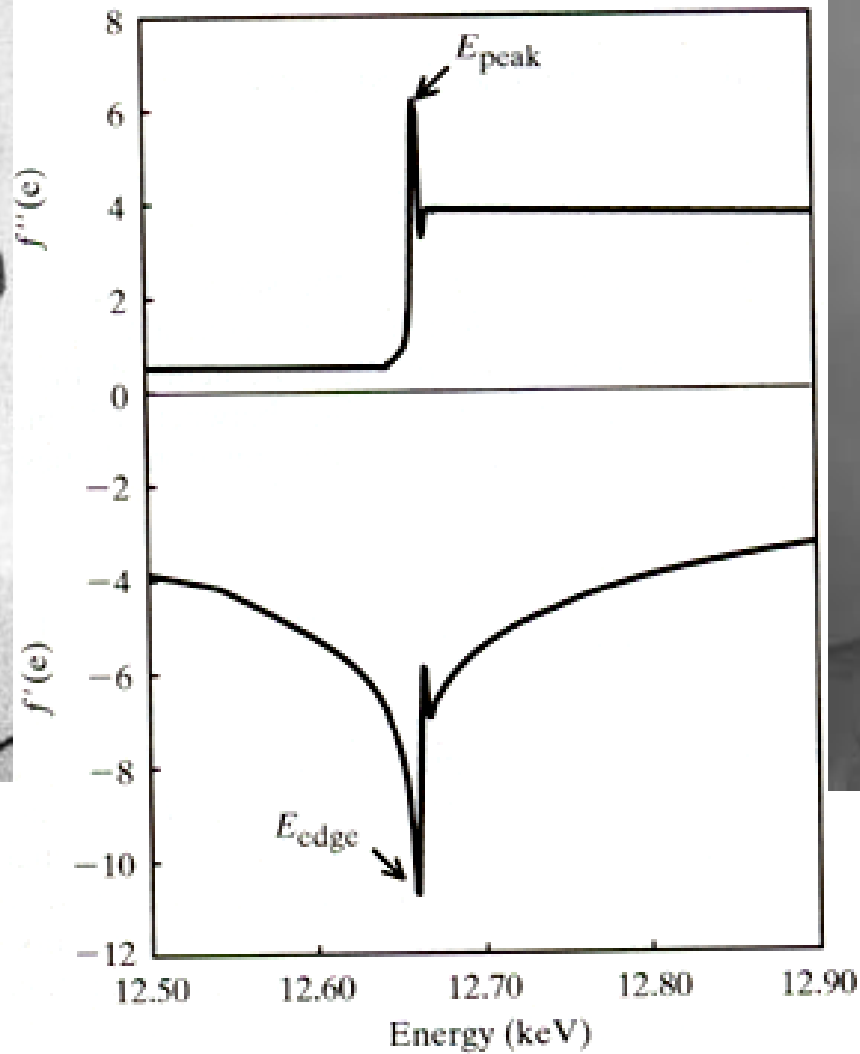


Figure 2

A typical experimental arrangement for a cryocrystallographic data collection. A magnetic rubber disc is pierced with a stainless steel pin, the top of which is rounded and the bottom of which fits into the hole in the goniometer head. The magnet must be strong enough to make a rigid connection, but weak enough to allow the experimenter fine control of the top hat, which is fabricated from a magnetic metal such as stainless steel or nickel. (Figure reproduced from Garman & Schneider, 1997.)

Counters, Films, multiwire, imaging plates, CCD





Selenium K-edge and peak for *E. coli* thioredoxin (Hendrickson et al)

Breakdown of Friedel's Law

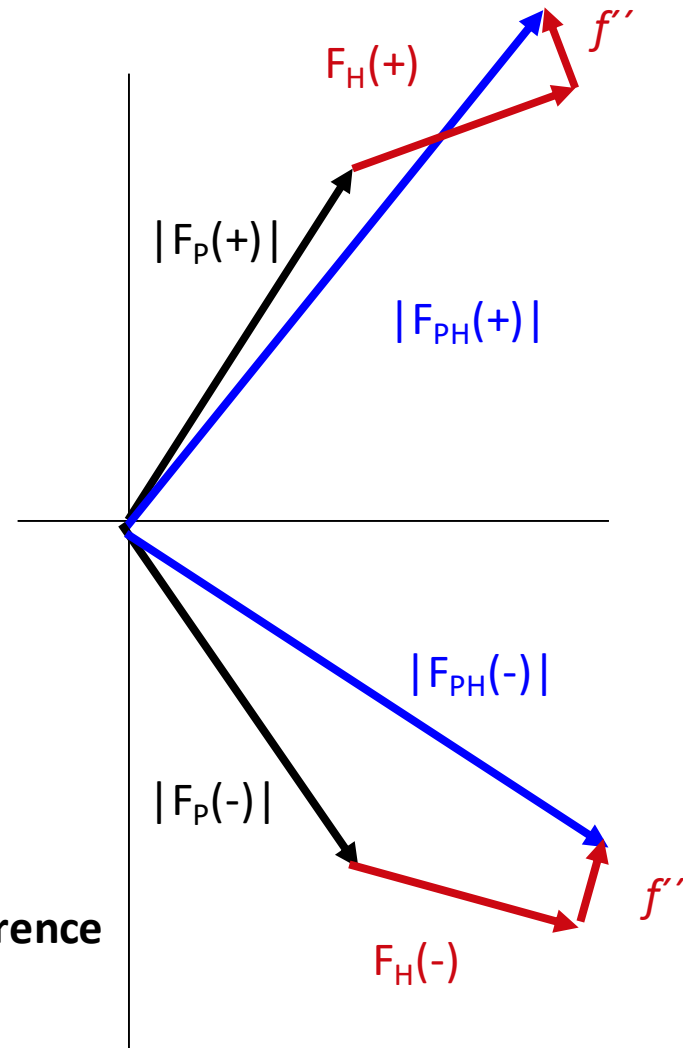
When an anomalous scatterer is present

$$f(\theta, \lambda) = f_0(\theta) + f'(\lambda) + if''(\lambda)$$

$$|F_{hkl}| \neq |F_{-h-k-l}|$$

We can measure this difference of amplitudes

$$\Delta F_{\pm} = |F_{PH(+)}| - |F_{PH(-)}| \text{ is the Bijvoet difference}$$



this can be used as an approximation to $|F_H|$ to locate sites

1980s-1990s



Molecular replacement

Can you recognize the similarity?

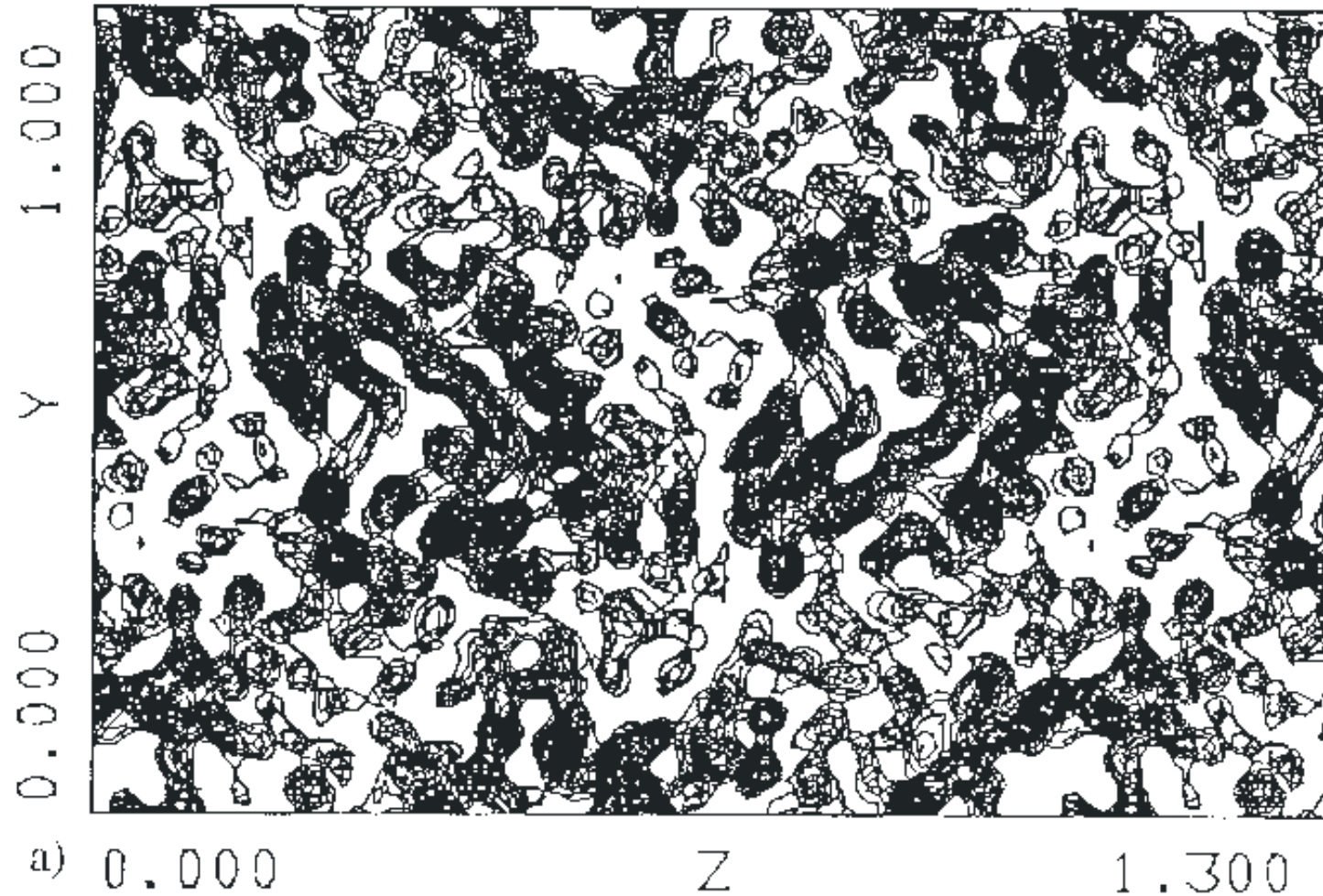


$$R(\kappa, \phi, \psi) = \int_{r_{\min}}^{r_{\max}} \mathbf{P}_{nat}(\mathbf{u}) \mathbf{P}_{mod}(\kappa, \phi, \psi, \mathbf{u}) d\mathbf{u}$$

$$\mathbf{T}(\mathbf{t}) = \int_{cell} \mathbf{P}_{2 \rightarrow 1}(\mathbf{u} - \mathbf{t}) \mathbf{P}_{nat}(\mathbf{u}) d\mathbf{u}$$

Density Modification

X = 0.69 - 0.81



Double sorting algorithm!

Perfecting the structure

$$f = \sum \frac{1}{\sigma^2} (|F_o| - |F_c|)^2$$

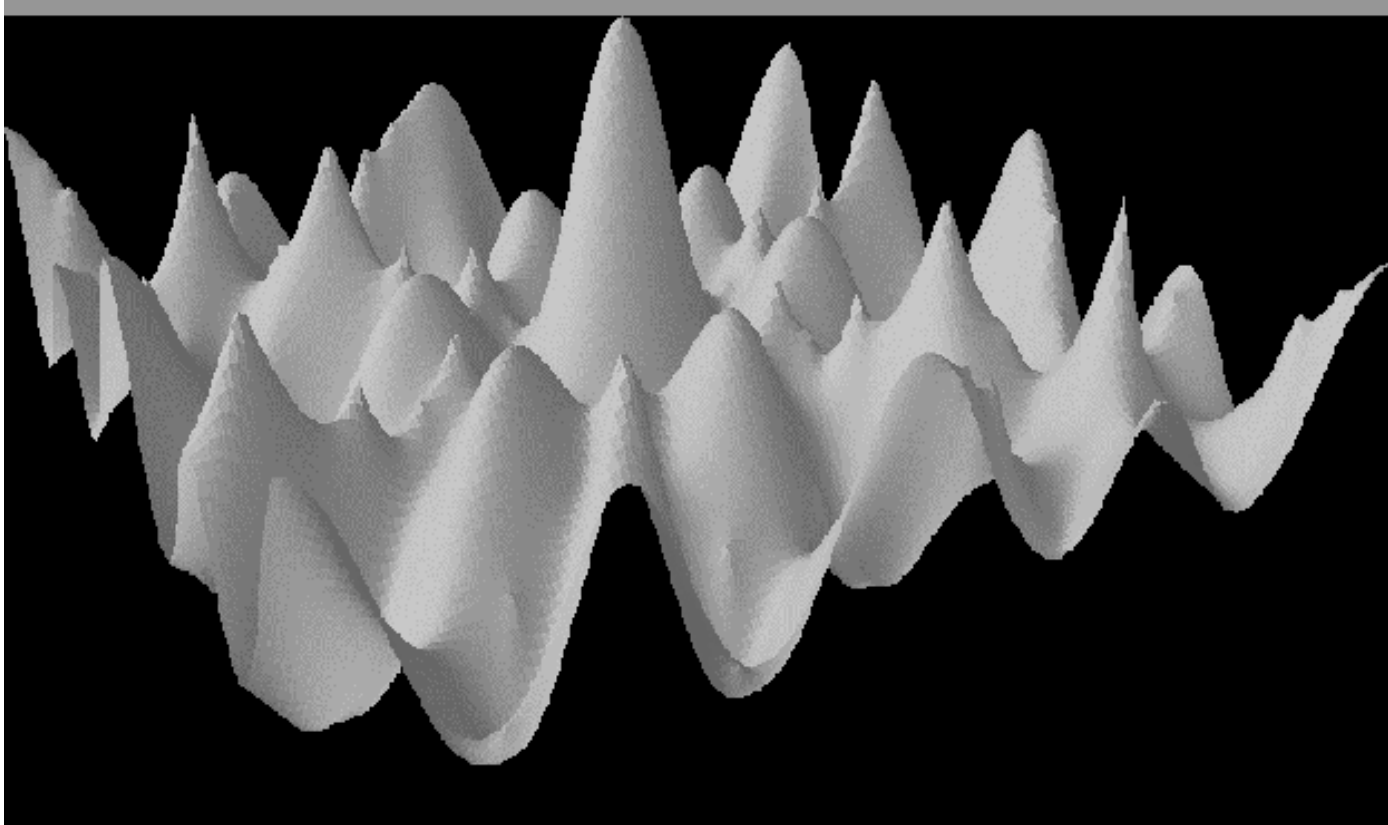
F_c s are calculated structure factors

$$F(hkl) = \iiint \rho(xyz) [\exp 2\pi i (hx + ky + lz)] dx dy dz$$

$$F(hkl) = \sum \sum \sum f(xyz) \exp 2\pi i (hx + ky + lz)$$

Importance of observations to parameters ratio

$$f = \sum \frac{1}{\sigma^2} (|F_o| - |F_c|)^2$$



NP problems: scary **local minima**

RESTRAINTS ARE THE SOLUTION

also...

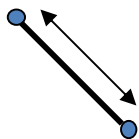
Planar groups

Hydrogen bonds

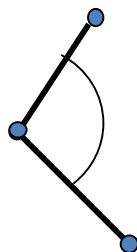
Energy

B's

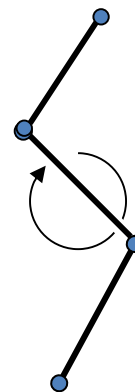
Stereochemical restraints



bond lengths



bond angles



torsion angles

Validation

Ramachandran plot

Bond lengths/angles

Homolog structure comparison

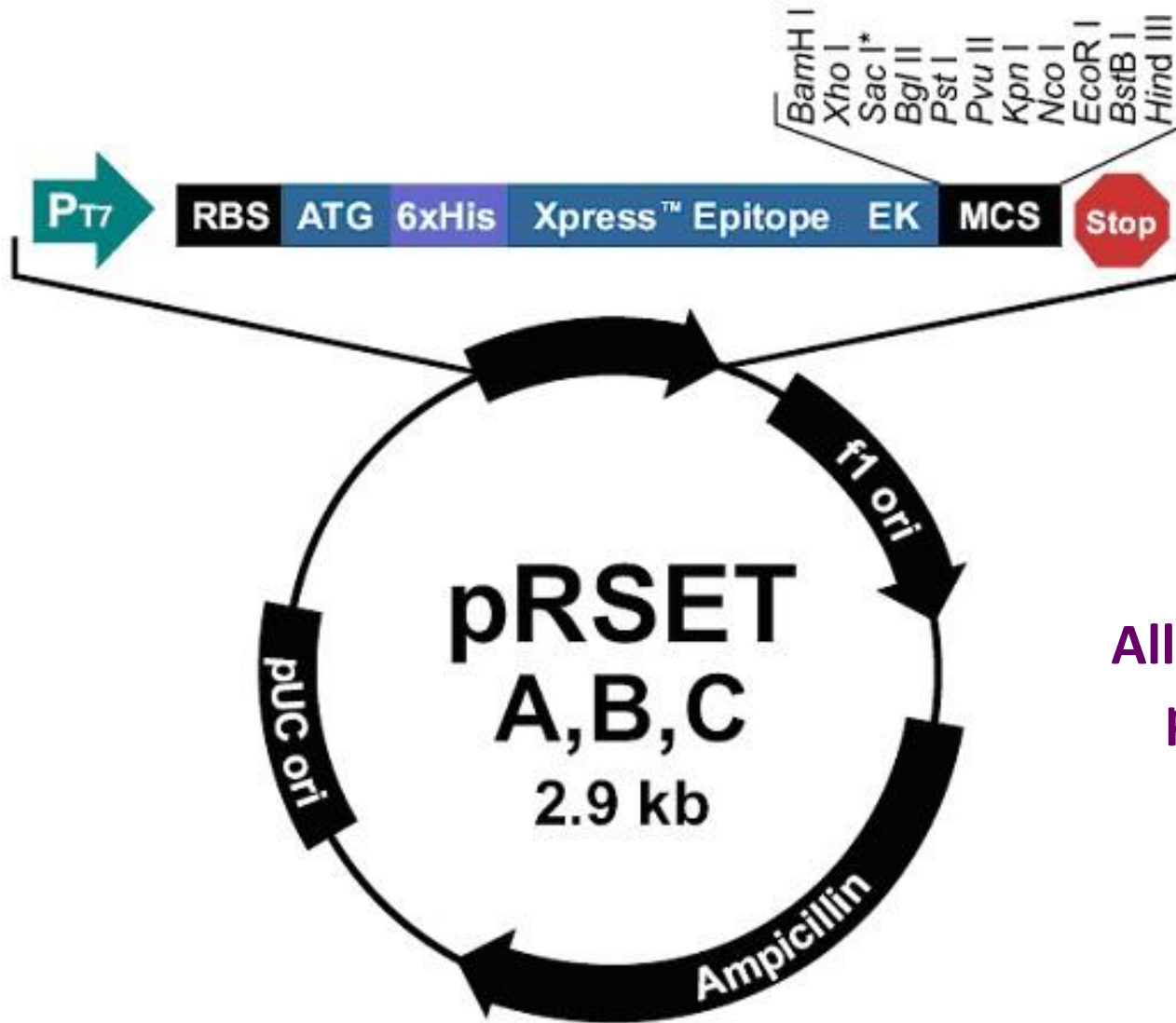
Independent structural solutions



Robots for setting up
crystallization and
examining results



pRSET vector



Allows one step purification

Most proteins are now obtained by bacterial or cell expression

Thank you



32 point groups (crystal classes)

Rotation axis only	1	2	3	4	6
Dihedral	222	32	422	622	
Isometric				23	432
	-	-	-	-	-
Rotoinversion axis	1	2	3	4	6
Rotation, \perp m	2/m	3/m	4/m	6/m	
Rotation, \parallel m	2mm	3mm	4mm	6mm	
		-	-	-	
Rotoinversion		32m	42m	62m	
Dihedral, m	2/m2/m2/m		4/m2/m2/m		6/m2/m2/m
	-		-		
Isometric, m	2/m3		4/m3m		