

the 44th crystallographic course at Ettore Majorana Centre, Erice, Italy June 2 to 12, 2011



Pushing the limits of Modern Powder Diffraction¹





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¹ As a reminder: "Yesterday today was tomorrow, but tomorrow today will be yesterday "

Content



- Instrumentation 1917
- Instrumentation 2011
- Exploring 2D powder patterns
- Parametric Rietveld refinement
- MEM + CF



The pioneers of powder diffraction 1916/1917





Peter J.W. Debye



Paul Scherrer



Albert W. Hull

OUTLINE OF METHOD. The method consists in sending a narrow beam of monochromatic Xrays (Fig. 2) through a disordered mass of small crystals of the substance to be investigated, and photographing the diffraction pattern produced. Disorder, as regards orientation of the small crystals, is essential. It is at-



Debye, P.; Scherrer, P. (1916), Interferenzen an regellos orientierten Teilchen im Roentgenlicht, Physikalische Zeitschrift 17, 277-283 Hull, A. W. (1917), A New Method of X-Ray Crystal Analysis, Phys. Rev. 10, 661 - 696

Crystal structure from `*in situ***' powder diffraction:** The beginning...



An early ,in-situ' powder xray diffraction experiment on the structure of α -N₂



Schematic drawing of the used cryo-camera



Debye-Scherrer film of α -N₂ at 34K Exposure time: 18h



The next big step: towards higher resolution







Guinier, A. (1937). Arrangement for obtaining intense diffraction diagrams of crystalline powders with monochromatic radiation. C. R. Acad. Sci. 204, 1115–1116.

The first "modern" powder diffractometer





Target Primary Soller Slit Divergence Slit Sample Anti-Scatter Soller Slit Soller Slit

Norelco powder, X-ray goniometers. First installed in 1949 at the Geophysical Laboratory (Design patented by William Parrish in 1947) Standard Bragg-Brentano geometry

Modern laboratory powder diffractometers





Bruker D8-Advance, Våntag PSD



Stoe-Stadi-P with 140° IP-PSD



PANalytical, X'pert PRO with Accelerator



Stoe-Stadi-P, 6° PSD's and Mythen

High speed, high resolution PSD's in Debye-Scherrer geometry



Inel CPS120 with 120° PSD

Modern powder diffractomters at synchrotron beamlines





Multi-analyzer crystals or 2D detectors in Debye-Scherrer geometry









Back to the roots ... Why Debye-Scherrer geometry in the laboratory ?



Advantages of the Debye-Scherrer method

- Less grain size effects
- Little preferred orientation
- Small amounts of material
- Handling of extremely sensitive samples
- Simple correction functions
- No overspill effect
- Simple line profile (\rightarrow fundamental paramters)
- Easy adaptation of reaction cells for non-ambient conditions, gas flow etc.
- Perfect for structure determination

Disadvantages of the Debye-Scherrer method

- Absorption (for Z > 20)
- High background
- Peak to background ratio

Solution → Using a high resolution Mo diffractometer with Ge(220) primary beam monochromator and a high efficiency detector

LaB₆ reference pattern of the Mo-K_{a1} radiation high resolution Debye-Scherrer diffractometer





 LaB_6 line profile standard measured with Mo-Ka1 radiation (Bruker D8 Advance with Ge(220) primary beam monochromator and Lynx-Eye detector with 0.5 mm thick solicon stripes, 17 h counting time)

4 refined parameters only

Resolution in 2 θ < 0.04° 2 θ Minimum d-spacing < 0.4 Å

Starting from an unknown chemical composition: a new nitridoosmate from lab data (Stoe-Stadi-P, Mo-K_{α1})



Known composition before simulated annealing: Ba/Os = 3:1 plus several nitrogen atoms



"Ba₃OsN₃"

R-wp 6.690 <u>R-Bragg 1.95</u>

R-wp 6.486 <u>R-Bragg 1.67</u>

 $(Ba_6O)(OsN_3)_2$

C. Schmidt, U. Wedig, R.E. Dinnebier, M. Jansen , Synthesis, Crystal structure, Properties and Bonding of Ba₆(O)(OsN₃)₂, (2008), Chem. Asian J. 3, 1983 – 1990

Alternatively: (Ba₆O)(OsN₃)₂



"Ba₃OsN₃" $(Ba_6O)(OsN_3)_2$ ●Os ●Ba ●N С С a a а

∎Os ∎Ba DN DO

Ba₃FeN₃ exists !

CRYSTAL06 (DFT)-Calculations – Hybrid-Funktional B3PW

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		exp.	calculated			
		(Ba ₆ O)(OsN ₃) ₂	(Ba ₆ □)(OsN ₃) ₂	(Ba ₆ O)(OsN ₃) ₂	(Ba ₆ N)(OsN ₃) ₂	(Ba ₆ O ₂) (OsN ₃) ₂
Position	3b	0	-	0	Ν	ο
	3a	-	-	-	-	О
Lattice parameter / Å	а	8.112	8.077	8.085	8.135	8.094
	С	17.390	17.348	17.414	17.599	17.359
Distance / Å	Ba –3b	2.73	2.82	2.71	2.70	2.80
	Ba –3a	2.91	2.82	2.92	2.96	2.80
Binding energy / (kJ / mole)	O / N in 3a / 3b			822	578	1586
Partial charges/	OsN ₃ -⁵		-2.8 / +1.9	-2.6 / <mark>+1.6</mark>	-2.4 / +1.1	-2.4 / +1.0
Spin charges	Ва		+1.2 / <mark>-0.1</mark>	+1.3 / <mark>+0.1</mark>	+1.4 / <mark>+0.1</mark>	+1.4 / 0
	Pos. 3b		-0.7 / -0.6	-1.9 / 0	-2.8 / 0	-1.8 / 0
	Pos. 3a		-0.7 / -0.6	-0.5 / +0.5	-0.5 / +0.4	-1.8 / 0

Calculations by Ulrich Wedig, MPI-FKF, Stuttgart

Data reduction & filtering: A high pressure phase of the NLO – compound BiB₃O₆ (BiBO)





- Better conversion efficiency than BBO (β-BaB₂O₄) or LBO (LiB₃O₅)
- Better resistance to laser damage than KTP
- Tunable laser 240nm 1000nm

R. E. Dinnebier, B. Hinrichsen, A. Lennie, and M. Jansen, The high pressure crystal structure of the NLO compound BiB₃O₆ from 2D powder diffraction data. (2009), Acta Cryst. B65, 1-10 (see also IUCr newsletter June 2009)

B. Hinrichsen, R. E. Dinnebier, M. Jansen, On the intensity distributions within Debye-Scherrer rings. What is different in high pressure experiments ? Part I: Theory & Part II: Application (2009), *. Kristallogr. Suppl. 30 (2009) 139-153*

Known phases of BiBO





Raw data at high pressure...







2D image plate powder diffraction data set of BiB_3O_6 at <u>high pressure</u>

How about counting statistics with 2D XRPD data ?



p

A very large number of equally sized and randomly oriented crystallites to the diffraction pattern would lead to an <u>ideal binomial intensity distribution</u> over the entire Bragg cone.

$$P_{B}(n|N) = {\binom{N}{n}} p^{n} q^{N-n}$$

$$= \frac{N!}{n!(N-n)!} p^{n} (1-p)^{N-n}$$
probability P_{B} of exactly n
successes out N trials where each
trial has the probability of success
and probability of failure $q=1-p$
 $n \to \infty$; $p = const$
Image plate

$$P_{N}(n) = \frac{1}{\sigma\sqrt{2\pi}} \exp\left[-\frac{(n-Np)^{2}}{2\sigma^{2}}\right]$$

$$P_{P}(n) = \frac{v^{n}e^{-v}}{n!}$$

Continuous Gaussian distribution

Discrete Poisson distribution

Intensity distribution in a bin









histogram of the intensities contributing to one bin* of air scattering intensity

The distribution of the air scattering intensity as a function of the azimuth

*A bin is a container into which pixels are grouped. It spans a 2D region of 2θ which is identical to the 2θ step width of the integrated pattern. The intensities of the pixels within a bin determine the corresponding intensity of a step in the integrated pattern.

The simpler the better ... Fractile filtering



We propose a robust type of **band pass** filter based on *fractile statistics*. A fraction x of the low intensity data and a fraction y of the high intensity data are removed:

$$I_{\min} + x (I_{\max} - I_{\min}) \ge I_{filtered} \ge I_{\min} + (1 - y) (I_{\max} - I_{\min})$$

Fractile filtering on ideal data





Effect of fractile filtering on the intensities of an almost ideal Bragg reflection as it would contribute to a two-theta bin.

The effect of filtering on high pressure XRPD data





The intensity is shown as a function of the azimuth for a high pressure powder diffraction data set (+) highest data fraction removed by the filter.

(-) lowest data fraction removed by the filter.



The azimuthal intensity distribution of a Bragg peak

Most promising description of the intensity distribution is the <u>Pareto distribution (80/20 rule</u>) convoluted with the normal distribution.

(At values of a>20, the Pareto distribution tends toward a Dirac delta function, thereby reducing a convoluted normal-Pareto (NP) function to the normal distribution of perfectly monodisperse grains)

Estimating the high intensity fraction to be filtered





For high values of parameter *a* the normal fraction approaches 1 asymptotically.

For a filter setting which would reduce the Pareto distribution to a roughly normal distribution (**1-normal fraction**) of the highest intensities should be removed.

Finally, the filtered data...





The effect of filtering on the diffraction image is shown in this figure. Intensities per integration bin:

Green	mask	\rightarrow
	mask	

21116	ma	c V
Juc	7 I I I a	SN

Yellow mask

 \rightarrow

bottom 2% beam stop

top 48%

Screenshot of Powder3D-IP http://www.fkf.mpg.de/xray/

The effect of filtering

Red: integrated pattern (left) and standard deviation (right) of the unfiltered image Blue: <u>48% of the highest</u> intensities and <u>2% of the lowest</u> intensities removed

Filtered diffraction patterns of BiBO at high pressure

Simulated Guinier-plot of BiB₃O₆ in the pressure range from 0 to 12 GPa (using Powder3D)

Structure determination of ε-BiBO (TOPAS)

The toolbox:

- 1. Global optimization (Simulated annealing)
- 2. Optimized cost function —
- 3. Modified weighting scheme
- 4. Anti-bumping penalties
- 5. Distance restraints
- 6. Occupancy merging

Cost $8+10^{6}$ $4+10^{6}$ $2+10^{6}$ 300 Torsion 200 100 0.2 0.4 0.5 Translation Translation

From: Radovan ČERNÝ, Chem. Met. Alloys 1(2008) 120-127,

3D hypersurface of the cost function for two parameters

$$\chi^{2} = \sum_{h} \sum_{k} (I_{h} - c |F_{h}|^{2}) (V^{-1})_{hk} (I_{k} - c |F_{k}|^{2})$$

W. I. F. David, On the equivalence of the Rietveld method and the correlated integrated intensities method in powder diffraction, J. Appl. Cryst. (2004). 37, 621-628.

How do the lone pairs react on pressure...

Parametric Rietveld refinement using spontaneous strain, symmetry modes and polyhedral tilting

as a function of external variables

Sequential ←→ parametric Rietveld refinement

Sequential Rietveld refinement: Each parameter in each pattern is refined individually

$$y_{calc}(2\theta_{pattern(1)}) = function(p_{(1),1}, p_{(1),2}, ..., p_{(1),m})$$

....
$$y_{calc}(2\theta_{pattern(n)}) = function(p_{(n),1}, p_{(n),1}, ..., p_{(n),1})$$

Minimization for each powder pattern independently

$$Min = \sum_{i} \left(w_i \left(y_{calc} \left(2\theta_i \right) - y_{obs} \left(2\theta_i \right) \right)^2 \right)$$

Parametric Rietveld refinement: Some parameters in each pattern are functions of external variable(s)

$$y_{calc}(2\theta_{pattern(1)}) = function(p_{(1),1}, p_2 = f(T_1, T_2, ..., T_t))..., p_{(1),m})$$

...
$$y_{calc}(2\theta_{pattern(n)}) = function(p_{(n),1}, p_{(n),2} = f(T_1, T_2, ..., T_t))..., p_{(n),1})$$

Minimization over all powder patterns simultaneously

$$Min = \sum_{pattern=1}^{n} \left(\sum_{i} \left(w_i \left(y_{calc} \left(2\theta_{pattern,i} \right) - y_{obs} \left(2\theta_{pattern,i} \right) \right)^2 \right) \right)$$

Advantages of "parametric Rietveld refinement"

The idea:

The evolution of parameters in Rietveld refinement is described by functions in dependence on external variables (temperature, time, pressure, ...). The parameters of these functions are treated as global parameters and are subjected to *"surface*" Rietveld refinement (all powder patterns are refined <u>simultaneously</u>).

Advantages:

- the correlation between parameters and the final standard uncertainty can be reduced
- physically meaningful constraints and restraints can be introduced easily
- non-crystallographic parameters can be refined (e. g rate constants, temperatures, ...)

Idea: Parameterization of the order parameter in dependence on temperature

The order parameter

Many crystalline solids can be considered as distorted versions of a higher symmetry parent structure. This parent structure can either be virtual or real. A group-subgroup relation must exist between the structures. All symmetry elements of the low-symmetry phase are already present in the high symmetry phase.

A new thermodynamic variable is necessary to specify the thermodynamic state of the low symmetry phase: order parameter η

V. K. Wadhawan, Introduction to Ferroic Materials, (Gordon and Breach, Singapore, 2000). E.K.H. Salje, Phase Transitions in Ferroelastic and Coelastic Crystals (Cambridge Univ. Press, Cambridge, 1993).

The order parameter in Landau theory

Landau theory is believed to decribe the main physical features of f<u>erroelastic</u> and <u>co-elastic</u> phase transitions.

Excess Gibbs (free) energy is the difference in Gibbs energy between both phases, stabilizing the low symmetry phase: $G_o = G_o(T, P, N, \eta)$

Equilibrium condition: $\partial G/\partial \eta = 0 \rightarrow G=G(T,P,N)$

Trivial solution G(T)=0 (for high temperature phase)

All quantities are measured with respect to the high temperature phase as excess quantities

Landau assumed that the excess free energy (Landau potential) can be described in a Taylor series for small values of η . Assumption: polynomial form of G is a good approximation over an extended temperature interval and for large values of η .

 $G = GO + \alpha \eta + A\eta 2 + C\eta 3 + B\eta 4 + \dots GO, \alpha, A, B, C = f(P,T) \qquad GO \neq f(\eta)$

Power law behavior of the order parameter

T/T_c

Temperature evaluation of structural order parameters η for several 2nd order and tricritical phase transitions (from Salje, 1993)

Access to the order parameter via spontaneous strain

Fig. 8. Spontaneous-strain data for the tetragonal \Rightarrow orthorhombic transition in As₂O₅ (data from Redfern & Salje, 1988; Redfern, 1988). (a) Lattice-parameter variation with temperature. (b) Variation of unit-cell volume with temperature, showing a small volume change due to the transition. The straight lines drawn in would imply $V_s \propto T$. (c) The square of the symmetry-adapted form of the symmetry-breaking strain is linear with T, implying $Q^2 \propto T$ and, hence, second-order behaviour.

Carpenter, M.A., Salje, E.K.H. and Graeme-Barber, A. (1998) Spontaneous strain as a determinant of thermodynamic properties for phase transitions in minerals. Eur. J. Mineral., 10, 621–91

Access to the order parameter via symmetry/distortion modes

distortion mode = Amplitude x polarization vector

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$$\mathbf{r}_{j}^{LS} = \mathbf{r}_{j}^{HS} + \sum_{m} c_{m} Q_{m} \boldsymbol{\varepsilon}(j|m)$$

Mois I. Aroyo, J. M. Perez-Mato and Danel Orobengoa, Universidad del Pais Vasco, Bilbao, Spain, THE BILBAO CRYSTALLOGRAPHIC SERVER: COMPUTER TOOLS FOR STRUCTURAL PHASE TRANSITIONS, lecture at "International School on the use and application of the Bilbao Crystallographic Server", 21-27 June 2009 Lekeitio (Spain)
The concept of polyhedral tilting, as an alternative way of describing distortions





Double-tetrahedra-group with 3 internal deg. of freedom as main structural building unit :

- average Fe-O bond length
- tilt1 torsion angle between the tetrahedra
- tilt2 rotation angle of one tetrahedron with respect to the other

The concept of polyhedral tilting: Hazen, R.M. and L.W.Finger (1982) Comparative Crystal Chemistry: Temperature, Pressure, Composition and the Variation of Crystal Structure. London: J. Wiley & Sons, xv, 231 p.

Example I: CsFeO₂(HT)





In-situ XRPD data of CsFeO₂, recorded at SLS with the 1D-Mythen-detector (4*10 seconds for each scan)

Structure of RbFeO₂, refined from a reticular pseudomerohedrally twinned crystal with six domains, J. Nuss, N. Z. Ali and M. Jansen, Acta Cryst. (2007). B63, 719-725

Aristotype: filled β-cristobalite structure







$CsFeO_2$ (HT)

β -Cristobalite

Phase transformation of CsFeO₂





Structural distortion mainly driven by

rotation of the Fe₂O₇ double tetrahedra & translation of Cs atoms

Distortion modes amplitudes of CsFeO₂



ISODISPLACE: view distortion Graphical rendering of selected crystal (help) 1.00 Master Slider cs Modes E cs_1 [0.09, 1.00] cs_2 [0.26, 1.00] -0.05 GM5+[cs:b]T2(a) 0.05 DT5[cs:b]T2 1(a) 0.05 DT5[cs:b1T2 2(a) -0.25 SM2[cs:b]T2(a) L3+fcs:b1T2(a) 0.00 X1[cs:b]T2(a) 0.00 GM1+fcs:blorder(a 0.00 X1[cs:b]order(a Fe Modes 1 [0.00, 1.00] Fe_2 [0.00, 1.00] 0.00 GM5+[Fe:a]T2(a) 0.00 DT5[Fe:a]T2 1(a) 0.00 DT5[Fe:a]T2 2(a) 0.00 SM2[Fe:a]T2(a) 0.00 L3+IFe:a1T2(a) 0.00 X1[Fe:a]T2(a) 0.00 GM1+[Fe:alorder(a) 0.00 X1[Fe:a]order(a) O Modes O 1 [0.63, 1.00] O 2 10.58, 1.001 O 3 [0.54, 1.00] 0 4 10.54, 1.001 0.00 SM2[O:c]A2u(a) 0.00 SM2IO:clEu 1(a) 32.0 fps 🖌 Atoms 🖌 Bonds 🖌 Cells 📃 Spin 📃 Animate Color Direction: 0 Apply View 🔘 Xrot Yrot Zrot Zoom SupHKL SupUVW ParHKL ParUVW

B. J. Campbell, H. T. Stokes, D. E. Tanner and D. M. Hatch, "ISODISPLACE: a web-based tool for exploring structural distortions", J. Appl. Cryst. (2006). 39, 607-614

B. J. Campbell, J. S. O. Evans, F. Perselli, H. T. Stokes, "Rietveld refinement of structural distortion-mode amplitudes",

IUCr Computing Commission Newsletter 8, 81-95 (2007).

M. Müller, R. E. Dinnebier, N. Z. Ali, B. J. Campbell and M. Jansen, Parameterized distortion modes versus rigid body Rietveld refinement, a case study of CsFeO₂, Materials Science Forum Vol. 651 (2010) pp 79-95.

(Main) distortion mode amplitudes of CsFeO₂



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Difficulty: some modes are coupled !

Distortion modes & amplitudes for CsFeO₂



1 Fd-3m[0,0,0]GM5+(a,0,0)[cs:b]T2(a) -0.0113(3)

2 Fd-3m[0,3/2,0]DT5(0,0,0,0,0,0,0,0,a,-0.414a,0.414a,a)[cs:b]T2_1(a) -0.0506(3)

3 Fd-3m[0,3/2,0]DT5(0,0,0,0,0,0,0,0,a,-0.414a,0.414a,a)[cs:b]T2_2(a) 0.1358(4

4 Fd-3m[1/2,1/2,0]SM2(0,a,0,0,0,0,0,0,0,0,0,0)[cs:b]T2(a) -0.2839(4

- 5 Fd-3m[1/2,1/2,1/2]L3+(0,0,0,0,a,-a,-a,a)[cs:b]T2(a) 0.0132(4)
- 6 Fd-3m[0,1,0]X1(0,a,0,0,0,0)[cs:b]T2(a) 0.0138(3)
- 7 Fd-3m[0,0,0]GM5+(a,0,0)[Fe:a]T2(a) -0.0020(5)
- 8 Fd-3m[0,3/2,0]DT5(0,0,0,0,0,0,0,0,a,-0.414a,0.414a,a)[Fe:a]T2_1(a) -0.1180(4)
- 9 Fd-3m[0,3/2,0]DT5(0,0,0,0,0,0,0,0,a,-0.414a,0.414a,a)[Fe:a]T2_2(a) 0.0168(4)

10 Fd-3m[1/2,1/2,0]SM2(0,a,0,0,0,0,0,0,0,0,0,0)[Fe:a]T2(a).

Multiple simultaneous Rietveld refinements

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Parameterised strain modes in dependence on temperature



Low symmetry orthorhombic phase (supercell)

$$e_{11s} = \frac{a_s}{a_{s0}} - 1 = \frac{a_s}{a_{p0} / \sqrt{2}} - 1$$
$$e_{22s} = \frac{b_s}{b_{s0}} - 1 = \frac{b_s}{\sqrt{2}a_{p0}} - 1$$
$$e_{33s} = \frac{c_s}{c_{s0}} = \frac{c_s}{2a_{p0}} - 1$$

High symmetry cubic phase

$$e_{11p} = e_{22p} = e_{33p} \text{ and } e_{12p} = e_{23p} = e_{13p} = 0$$

$$e_{11p} = e_{22p} = \varepsilon_{\Gamma_1^+} - \frac{1}{2}\varepsilon_{\Gamma_3^+} = \frac{a_p}{a_{p0}} - 1 = \frac{b_p}{a_{p0}} - 1$$

$$e_{33p} = \varepsilon_{\Gamma_1^+} + \varepsilon_{\Gamma_3^+} = \frac{c_p}{a_{p0}} - 1$$

$$e_{12p} = e_{21p} = \frac{1}{2}\varepsilon_{\Gamma_5^+} = \frac{\pi}{2} - \gamma_p$$

$$e_{11s} = e_{11p} + e_{12p}$$
$$e_{22s} = e_{11p} - e_{12p}$$
$$e_{33s} = e_{33p}$$

relationship between the strain of the supercell and the cubic strain

If
$$(T < T_{crit})$$
 then $\varepsilon_{\Gamma}(T) = f_{\Gamma}(T_{crit} - T)^{\beta_{\Gamma}}$, else $\varepsilon_{\Gamma} = 0$

Parameterised strain modes in dependence on temperature



Temperature-dependent supercell parameters for CsFeO₂ as calculated from parametrically-refined power-law models of the strain parameters.

Temperature-dependent supercell strains for CsFeO₂ as calculated from their parametrically-refined power-law models



Order parameter η **derived from parametric Rietveld refinement**





Parameterised distortion modes versus polyhedral tilting in dependence on temperature

Order parameter η derived from parametric Rietveld refinement



distortion mode	critical exponent	rigid body	critical exponent
a3	0.27 (1)	Cs 1 y	0.21(1)
a4	0.08(1)	Cs 2 x	0.08(1)
a10	0.11 (1)	tilt-1	0.125(2)
a15 / a19	0.014(1)	tilt-2	0.009(1)
ε ₁	0.125(6)	ε ₁	0.12(1)
ε ₂	0.03(1)	ε ₂	0.03(1)
ε ₃	0.16(1)	ε ₃	0.15(1)
ε _S	0.23(1)		0.23(1)

$\eta = A(T_c - T)^{\beta}$	
$\beta = [0.070.125]$	

Some results:

- order parameters derived from polyhedral tilting and distortion modes are equivalent.
- The spontaneous strain e_s is coupled linear-quadratic with the a10 distortion mode and the angle tilt-1: $e_s \sim \eta^2$
- The spontaneous strain e_s is coupled linear-cubic for a4 and Cs2x : $e_s \sim \eta^3$
- The phase transition is clearly of first order.

Example II: As₂O₅ (HP)



330

$$a = \sqrt[3]{V(P)} = \left[V_{0a} \left(1 + \frac{K'_{a}P}{K_{0a}} \right) \right]^{-\frac{1}{3K'_{a}}}$$

1,02 20 1,01 а parameters 18 1,00 Murnanghan 16 0,99 C Pressure (GPa) 14 0.98 12 ell B 0,97 10 Normalised 0,96 0,95 0,94 0,93 8 6 2 0.92 280 290 300 310 320 12 16 18 2 10 Volume (A³) Pressure (GPa)

"Linear" Murnagham parameterisation of lattice parameters in dependence on high pressure. Comparison of relative unit cell parameters in dependence on pressure obtained by independent refinement (red) and by Murnaghan parameterisation (black).

Resulting EoS from parameterised lattice parameters

Ivan Halasz, Robert E. Dinnebier, and Ross Angel, "Advantages of using parametric Rietveld refinement for the evaluation of powder diffraction patterns collected in dependence on pressure.", 2010, J. Appl. Cryst.

As₂O₅ (HP) polynomial parameterization of coordinates

 $a = A_0 + A_1 P + A_2 P^2 + A_3 \sqrt{P}$



Joint plot of all the powder patters of As₂O₅ in dependence o pressure refined with polynomial parameterisation of coordinates



Overlapped projections of the crystal structures of As_2O_5 obtained by (red) independent refinement and (blue) refinement using polynomial parameterisation of coordinates at the pressure of 20 GPa.

Example III: CulnSe₂





phase transition is driven by:

- 1) anti site ordering of copper and indium atoms
- 2) translation of slenium atoms



S. Schorr, G. Geandier, Cryst. Res. Technol., 2006 41(5), 450-467.

The modes





three different types of modes: strain, displacive, occupancy

ISODISTORT: Campbell B. J., Stokes H. T., Tanner D. E. and Hatch D. M. (2006). J. Appl. Cryst. **39**, 607-614 Subgroupgraph: S. Ivantchev, E. Kroumova, G. Madariaga, J. M. Perez-Mato, M. I. Aroyo, *J. Appl. Cryst.* **2000**, 33, 1190-1191.

Parameterization of strain, displacive, and occupancy modes







Some results of parametric refinement



mode	critical exponent	
a1	0.24(1)	tricritical displacive phase transition
<i>s1</i>	0.49(2)	linear quadratic coupling
<i>s2</i>	0.16(5)	maybe order-disorder transitions?
occupancy	0.28(1)	3D ordering phenomenon (Ising model)
temperature factor	critical exponent	
cations	0.24(1)	same as <i>a1</i>
anion	0.49(2)	same as <i>s1</i>

 p phase transition combines an order-disorder (cation lattice) and a displacive (anion lattice) transition

Towards automation: Powder 3D¹ - Parametric





¹ B.Hinrichsen, R.E.Dinnebier and M. Jansen, *Z. Krist,* (2004) **23**, 231-236. ² .G.Stinton , J. Evans, *J. Appl. Cryst,* (2007) **40**, 87-95

Handling TOPAS input files with Powder3D-Parametric





Sequential refinements with Powder3D-Parametric





Parametric refinements of CuPc (isothermal

with Powder3D-Parametric



00% alpha and 10% beta CUPC at 250°C



Avrami paramameters for parametric refinements:

sequential refinements for comparison:

n = 0.82163 (1-130) k = 0.00227 n = 0.92622 (50-130) k =0.00213

n = 0.97 k = 0.00178

"True" disorder from XRPD by MEM and charge flipping



The method of MEM

$$S = -\sum_{j=1}^{N} \rho_j \log\left(\frac{\rho_j}{\omega_j}\right)$$

- number of pixels in the unit cell;
- ρ electron density in pixel *j*;
 - initial value of the electron density or prior

Collins (1982)

Entropy:

Ν

Q

The method of charge flipping with histogram matching



Oszlányi and Sütő Acta Cryst. (2004). A60, 134-141 Baerlocher, McCusker and Palatinus *Z.Krist.* (2007). **222** 47-53

The high temperature phases of rubidium oxalate by *in situ* powder diffraction





R. E. Dinnebier, S. Vensky, M. Jansen, and J. Hanson, Crystal Structures of and Topological Aspects on the High Temperature Phases and the Decomposition Products of the Alkali Oxalates $M_2[C_2O_4]$, M=(K, Rb, Cs), 2005, Chemistry, a European Journal, 11, 1119 – 1129.

A. Samy, R. E. Dinnebier, S. van Smaalen, and M. Jansen, The Maximum Entropy Method and Charge Flipping, a powerful combination to visualize the true nature of structural disorder from *in situ* X-ray powder diffraction data. *(2010) Acta Cryt. B.*

Crystal structures of the different phases of Rb₂C₂O₄ and Rb₂CO₃





Example: Rietveld plot of a-Rb₂C₂O₄









Rietveld refinement : GOF= 1.57, R_p =1.70, R_{wp} =2.77

Example: LeBail plot of a-Rb₂C₂O₄







LeBail: GOF= 0.59, $R_p=0.74$, $R_{wp}=1.03$

Observed structure factors & model-biased effects





Rietveld refinement \rightarrow F_{obs} -model-bias

$$A(obs)_1 = \sum_{i=1}^n \frac{A(calc)_1 \times q_1(i)}{A(calc)_1 \times q_1(i) + A(calc)_2 \times q_2(i)} (obs(i) - back(i))$$

$$A(obs)_{2} = \sum_{i=1}^{n} \frac{A(calc)_{2} \times q_{2}(i)}{A(calc)_{1} \times q_{1}(i) + A(calc)_{2} \times q_{2}(i)} (obs(i) - back(i))$$

Poulor profile based on prf file

LeBail algorithm

$$A_{m}^{r+1}(obs) = \sum_{i=1}^{N} \frac{A_{m}^{r}(obs) \times q_{m}(i)}{\sum_{m=1}^{N} A_{n}^{r}(obs) \times q_{n}(i)} (obs(i) - back(i)), \text{ where, } A_{m}^{r=1} = 1, n = 1, \dots, N$$

Ways of reconstructing the different types of electron density maps.





MEM algorithm: Sakata&Sato (1990); Computer programs: BayMEM, Jana, Superflip

Results of MEM-calculations based on model-Rietveld-refinement





(c1)

(c2)

Results of the combination of MEM with phases of CF





Not perfect, but a powerful combination to solve disordered crystal structures from powder data directly...

My view of the future (it starts now...)



- Many experiments which were only be possible at the synchrotron can now be done in the laboratory (e.g. Mo K_{$\alpha1$} in Debye-Scherrer geometry)
- Future Rietveld-programs: toolboxes, macros, self learning, user community
- Determination from powder data is becoming routine. (for small and medium sized structures).
- Parametric Rietveld refinement opens a new world (determination of non-structural parameters, e.g. kinetics, order parameters)
- Accurate electron density distributions using MEM method from XRPD
- Complex structure determination from powder data by the combination of CF and MEM.
- High speed (<1 s) + high resolution is now available at the synchrotron (ESRF, SLS, Petra III, Diamond, etc.)
- High pressure powder diffraction and PDF analysis in the laboratory....

Powder Diffraction Theory and Practice



Acknowledgement



To where the money came from MPG, FCI, BMFT To the facilities providing beamtime NSLS, ANKA, ESRF, APS To collaborators

Peter Stephens (SUNY at Stony Brook), John Hanson (NSLS), Sander van Smaalen (Bayreuth), Bernd Hasse (Incoatec), Ralph Weigel (ANKA), Andy Fitch (ESRF), Haozhe Liu (APS), Branton Campbell (BYU) ...

Fo current and former group members

















Constraints in MEM



1. Normalization of ρ to the expected number of electrons per unit cell volume (F_{000})

$$\int_{V} \rho dV - N_{el} = 0$$

2. F-constraints C_F with central moment *mn*. The resulting error distribution must be obeying the Gaussian distribution function

$$C_{F_n} = -1 + \frac{1}{m_n(Gauss)} \frac{1}{N_F} \sum_{i=1}^{N_F} \left(\frac{F_{obs}(\overrightarrow{H_i}) - F_{MEM}(\overrightarrow{H_i})}{\sigma(F_{obs}(\overrightarrow{H_i}))} \right)^n$$

3. G-constraints, sometimes phases or even amplitudes of individual structure factors cannot be determined reliably. This is often the case for powder diffraction data, where systematic and/or accidential overlap of reflections is common.

$$G_{G} = -1 + \frac{1}{N_{G}} \sum_{i=1}^{N_{G}} \left(\frac{G_{obs}^{i} - G_{MEM}^{i}}{\sigma(G_{obs}^{i})} \right)^{2} \qquad G^{i} = \sqrt{\sum_{j=1}^{N_{g}^{i}} \left[\frac{m_{j}}{\sum m_{j}} \left| F(\overline{H_{j}}) \right|^{2} \right]^{2}}$$

Sakata & Takata, 1996

Basics of Maximum Entropy Methode (MEM)



Maximize:

$$Q(\rho) = S(\rho) - \sum_{j=1}^{N_c} \lambda_j C_j(\rho), \longrightarrow \frac{\partial Q}{\partial \rho_i} = 0$$

$$\rho_{i} = \frac{N_{el}N_{pix}}{V}\tau_{i} \exp\left(-\lambda_{F}\frac{\partial C_{F}}{\partial \rho_{i}}\right) / \sum_{i=1}^{N_{pix}}\tau_{i} \exp\left(-\lambda_{F}\frac{\partial C_{F}}{\partial \rho_{i}}\right)$$

 $\tau_i^{n+1} \approx \rho_i^n$

αππροξιματιονσ

(3)	Set of N _{pix} nonlinear
(-)	equations

Electron density ρ_i Prior density τ_i N_{pix} no of pixels of cell volume V N_{el} no of electron/unit cell = F000 weight factor w standered error of F_{obs} $\sigma(H)$ ■ F_{obs} observed structure factors **F**_{MEM} MEM structure factors Lagrange multipliers λ no of constraints Nc

Iteration:

$$\rho_i^{n+1} = \frac{N_{el}N_{pix}}{V} \rho_i^{(n)} \exp\left(-\lambda_F \frac{\partial C_F}{\partial \rho_i}\Big|_{\rho_i^{(n)}}\right) / \sum_i \rho_i^{(n)} \exp\left(-\lambda_F \frac{\partial C_F}{\partial \rho_i}\Big|_{\rho_i^{(n)}}\right)$$
Sakata & Sato algorithm (1990)

The iteration is started with $\rho_i^{(1)} = \tau_i$ and the new density $\rho_i^{(n+1)}$ is calculated from the prior density $\rho_i^{(n)}$, the value of the constraint decreases each cycle untile the condition of $C_F \le 1$ is fulfilled
The parameters of MEM-calculations



Chemical formula	\Box -K ₂ C ₂ O ₄	\Box -Rb ₂ C ₂ O ₄	\Box -Rb ₂ CO ₃
Number of voxels	108 🗆 64 🗆 36	96 🗆 96 🗆 108	54 🗆 54 🗆 72
Pixel size /Å	0.101 0.095 0.096	0.067□ 0.067 □0.076	0.109□0.109□0.10 8
Electrons per unit cell	164	236	208
Lagrange multiplier	0.05	0.05	automated
Initial density	Flat-prior	Flat-prior	Flat-prior
F-constraints	86	46	42
G-constraints	9	6	4

Initial density (flat-prior)= N_{el} / N_{pix} Algorithm: Sakata&Sato (1990)

Computer program: BayMEM (van Smaalen, 2003)

Simulated annealing and disorder





CsCp* : perfect fit in A_1/amd (a = 7.54Å, and c = 19.97 Å), but problems in solving the crystal structure (Cs at 0 0 0, Cp* (as pseudo atom) at 0 1/4 1/8 ???)

U. Behrens, R. E. Dinnebier, S. Neander, F. Olbrich, Solid State Structures of base free rubidium and cesium pentamethylcyclopentadienides. Determination by high Resolution Powder Diffraction, 2008, Organometallics. 27(20), 5398-5400.

10 years after the measurement







Structure determiantion "by hand" in maximal translationengleiche subgroup I2/d (C2/c)



Optimum results for bent Cp*-Cs-Cp* chains with two-fold rotational disorder of the Cp* rings

Why tetragonal ?





Most likely stacking faults perpendicular to the *ac*-plane lead to pseudotetragonal appearance (supported by anisotropic microstrain broadening with the maximum "strain" occurring long the vector $a\pm c$ and shape of thermal ellipsoids)

Temperature dependence of the order parameter for 2nd order phase transitions





G is continuous at phase transition

$$\rightarrow \cdot G/\cdot T$$
 has a kink at T = T_c

 $T > T_c : \eta \rightarrow 0$ \rightarrow Minimum of free energy at $\eta = 0 \rightarrow \cdot = 0$; A > 0

 $T < T_c : \eta > 0$ \rightarrow Minimum of free energy at $\eta \neq 0 \rightarrow \cdot = 0$; A < 0, B(P,T) > 0

Simplest choice: $A(P,T) = a (T-T_c)$

Assumption: C = 0

 \rightarrow G = G₀ + a(T-T_c) η^2 + B η^4 (2-4 potential)

 $\Rightarrow \cdot G/\cdot \eta = 0 \Rightarrow 2a(T-T_c)\eta = -4B\eta^3 \Rightarrow \eta^2 = a/(2B)(T_c-T) \Rightarrow \eta = c (T_c-T)^{1/2} \quad (T < T_c)$

Temperature dependence of the order parameter for "weakly 1st order" phase transitions





Assumption: C = 0Assumption: $B(P,T_c) < 0$

 $C = 0 \rightarrow G(P,T) = A(P,T)\eta^2 + B(P,T)\eta^4$

 → higher order terms are necessary to allow for a minimum of the free energy at any finite value of η
G = Aη² + Bη⁴ + Dη⁶ with D > 0

Assumption: B = 0

 \rightarrow G = G₀ + a(T-T_c) η^2 + D η^6 (2-6 potential)

$$\rightarrow \cdot G/\cdot \eta = 0 \rightarrow 2a(T-T_c)\eta = -6D\eta^5 \quad \rightarrow \eta^4 = a/(3D)(T_c-T) \rightarrow \eta = c (T_c-T)^{1/4}$$

At the point where B(P,T) changes sign, a 2nd order phase transition goes over to a 1st order phase transition (tri-critical point)