75 Years of the Powder Diffraction File[™]: An Overview of Database Development for Materials Characterization

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International Centre for Diffraction Data

A non-profit scientific organization dedicated to collecting, editing, publishing, and distributing powder diffraction data for the identification of materials. The membership of the ICDD consists of worldwide representation from academe, government, and industry

Mission - The International Centre for Diffraction Data will continue to be the world center for quality diffraction and related data to meet the needs of the technical community. ICDD promotes the application of materials characterization methods in science and technology by providing forums for the exchange of ideas and information



Phase Identification





Hanawalt Method

Collect powder patterns

Reduced data to d-spacings (peaks) and Intensities (series of xy data)

Use the strongest 3 lines and permute their order in an index

Match unknowns with the index then match the full pattern

d	20.0	9.9	2.67	$\begin{array}{c} d \text{ in } A \\ \lambda = .708 \end{array}$	$\frac{\mathbf{I}}{\mathbf{I}_{i}}$	$\begin{array}{c} d \text{ in } A \\ \lambda = .708 \end{array}$	$\frac{\mathbf{I}}{\mathbf{I}_{i}}$
<u>I</u>	1.00	0.18	0.10	10.0 9.9 3,40	1.00 0.19 0.05		
I	40	7	4	\$.22 3.15	0.03		
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-			Z =	1.61 1.57 1.32 1.330 1.240	0.05 0.03 0.03 0.03 0.03		
$a_{\circ} \equiv$ A = $D \equiv$	b. =	с С =	. =	1.190	0.05		
n ==	ω=	8	=	- 1996	1.44	12.8%	4



1938 Ludo Frevel, Don Hanawalt and Sid Rinn

> Chemical Analysis by X-ray Diffraction



1941



Card Files and Indexes transported from Midland, Michigan to ASTM headquarters in Philadelphia in Don Hanawalt's car, copied by hand – Set # 1 of the Powder Diffraction File

Joint Committee on Powder Diffraction Standard (JCPDS) –

1941, 1945

(ASTM Committee on X-ray and Electron Diffraction, British Institute of Physics, National Research Council)

Davey (Hull)	
Pickett, Wyman	
Hanawalt	
Wilson, Lawrence Bragg	
Barton	
Fink	
Fuller	
Bannister	
Huggins	
Boldyrev	
Harcourt	
Kerr	
Nelson	
Magos	
Richmond	

General Electric (then Penn State) General Electric The Dow Chemical Co. (Mg) British Institute of Physics Institute of Physics (USA) Aluminum Company of America New Jersey Zinc British Museum, Cambridge University Eastman Kodak Co. Institute of Mines, Leningrad, Russia American Mineralogist Columbia University Batelle Crane Co. US War Department

The Powder Diffraction File







1987 CD-ROM <u><1GB</u> 46,000 Entries









Database Overview

- Quality Marks
- Database Status of a given entry
- Subfile Classification
- Database Filters
- Structural Classification
- Raw Data Archival



Data Source









Quality Marks (QM)

- Upcoming release of Powder Diffraction File will have 848,757 entries (Inorganic+Organic)
- Quality marks are essential while working with larger database with similar diffraction patterns
- Data validation and the quality mark assignments is the most important step in the editorial process
- Every entry in Powder Diffraction File has quality mark
- If an entry does not meet the condition for star pattern, editorial comment(s) describing the reason will be added



QM for Experimental Patterns

- Star (Well characterized chemically and crystallographyically, No unindexed lines, Δ2θ≤0.03^o)
- **R** (d values from whole pattern fitting, like Rietveld, Le Bail refinement)
- I (Well characterized chemically No unindexed strong lines,Δ2Θ≤0.06^o)
- **B** (Do not meet the criteria for *, I)
- **O** (Poorly characterized, with editorial comment explaining the reason)
- **C** (author calculated d values)
- **H** (Hypothetical)
- M (Minimally acceptable non crystalline pattern)
- G (Good non crystalline pattern, usually has additional characterization other than XRD)



QM for Experimental Patterns

- Each and every pattern editorially reviewed including visual comparison of raw data vs d-I list
- Undergoes more than 100 data validation checks before gets published



Calculated Patterns QM

If we know the crystal structure, we can calculate the diffraction pattern using the equation

It is extremely important to make sure that the crystal structure used for the calculation is correct. In fact it is the rate determining step in the editorial process





Database status

- When multiple entries are available, database status is assigned editorially to avoid nearly duplicate result set
- Primary
 - Primary pattern for a given phase usually best quality, room temperature data
- Alternate
 - Alternate to primary. Not necessarily mean of poor quality
- Deleted
 - Unresolved errors
 - Duplicated entry

🗠 Environment	🗸 Status	🚖 Quality Mark
Ambient	Primary	Star
Press. (Non-ambient)	Alternate	Rietveld
Temp (Non-ambient)	Deleted	● Good 📃
	Deleted	Indexed
Press. & Temp. (Non-ambient)		
🔲 Atomic Coordinates 🗱	Minimal Accepta	
📃 Raw Diffraction Data 📸		 ▲ Ⅲ ▶







PDF databases: subfiles

Comprehensive data collections - classified and edited by experts





Subfiles

- Classified into various categories based on chemistry, properties and application
- Very useful in minimizing false positives while performing search/match using a large database
- Each subfile is defined by set of rules approved by the concerned subcommittee
- Subcommittee members review the file and continuously improvise it
- Library database gets updated frequently based on subcommittee's input
- Examples: Cement, Pharmaceutical, Ion Conductor, Forensic etc



Application of subfile in search/match



Without subfile filter False positive

Correct answer using subfile filter



Structural Classifications

- Pearson Symbol Code (PSC)
- Traditional Structure Type Notations
- ANX Formula
- Long descriptive Notation following Parthe's method
- Mineral Classification
- Zeolite Classification



Pearson Symbol Code







Pearson Symbol Code Types

✤Filled

 Do not distinguish partially occupied site from fully occupied one
 Example cP5

Actual

- Actual site occupation (true to the composition)
 Example cP4.50
- Without hydrogen



Ca0.5 Ta O3

Atom	Num	Wyckoff	Symmetry	x	у	Z	SOF	IDP	AET
Cu	1	1a	m-3m	0.0	0.0	0.0	0.5		12-b
Ta	2	1b	m-3m	0.5	0.5	0.5	1.0		6-a
0	3	3c	4/mm.	0.0	0.5	0.5	1.0		2#b



ANX Formula

ANX formula is a type of structural classification based on type of atoms (cations, anions) and their site occupation

- Coordinate of all the atoms in the structure must be defined to apply this classification successfully
- Atoms occupying same site are treated as one single atom type
- All the sites occupied by the same atom type is combined except when oxidation state is different

○Symbols used : Cations (A-M), Neutral (N-R), anions (X,Y,Z, S-W)

- Example
 - Ca Ti O3 is of ABX3 type
 - Fe3O4 is of AB2X4 type



Long descriptive Notation

- Structure Type Formula, Pearson Symbol, Space group number
- Examples: Cu3 Au,cP4,221; Cu3 As,cI64,220; NaCl,cF8,225



Structure type requirements

- Must crystallize in the same space group
- Similar cell parameter ratio
- Same Wyckoff positions (Wyckoff sequence)
- When all the above 3 conditions satisfy, we have similar atomic environments





Standardization of Crystal Structure

- First proposed by Parthe and Gelato (*Acta Cryst. (1984), A40, 169-183*)
- The method calculated standardization parameter $\tau\,$ based on
 - Shift of origin
 - Rotation of the coordinate system
 - Inversion of the basis vectors

The smallest $\boldsymbol{\tau}$ represents the standard description

$$\Gamma = \sum_{i=1}^{N} \left(x_i^2 + y_i^2 + z_i^2 \right)^{1/2}$$

N is number of atom sites, x,y,z are fractional coordinates



Why standardized data for comparison?

```
CeCu2, Imma, a=4.425,b=7.057, c=7.475
```

Cu (8h) 0 0.051 0.1648

Ce (4e) 0 0.25 0.5377

Standardization

KHg2, Imma, a=8.10,b=5.16, c=8.77 Hg (8i) 0.190 0.25 0.087

K (4e) 0 0.25 0.703

Standardization

```
CeCu2, Imma, a=4.425,b=7.057,
c=7.475
```

Cu (8h) 0 0.051 0.1648 Ce (4e) 0 0.25 0.5377 KHg2, Imma, a=5.16,b=8.10, c=8.77 Hg (8h) 0 0.06 0.163 K (4e) 0 0.25 0.547



Structure Type Applications

- Deriving Starting Model for Rietveld Refinements
 - Traditionally starting model for Rietveld refinements was developed based on chemical/crystallographic intuition. In other words it is individual's knowledge of structure types
 - Databases with Structure Type information can make seminal contributions to this effort. Powder Diffraction File in particular is advantageous as one can perform search match based on their diffraction pattern to explore the possible models.



Structure Type Applications

- Structural chemistry information
 - Database search results can also be sorted based on their structural chemistry
 - Can be used to explore the database from the materials design point of view
 - 3D crystalline structure related properties
 - Ferrolectric, Piezoelectric, Non Linear Optics, Transport



Mineral Classifications

- Related by partial structural similarities
- Examples: Double chain (Amphiboles), Sheet (Mica)
- Collection within the families based on specific similarties
- Homologus (humite), Homeotype (perovskite)
- Highest symmetry phase where structure arrangement remains unchanged
- Example: Diamond
- Composed of isostructural phases
- cF8 NaCl group

Famil

Subfamily

Supergroup

Group

Subgroup

• Subsets of group based on chemistry (halides, oxides etc)

Advances in X-ray Analysis **41**, 606-613 (1997)





Supergroup, Diamond cF8 Fd-3m



Group, Sphalerite (ZnS) cF8 F-43m



Subgroup, Chalcopyrite (CuFeS2)



Search Based on Mineral Classification

Ш

Mineral Classification 🔻

STB - Stibnite (Supergroup)
 STL - Stipnomelane (Family)
 SOH - Stottite (Supergroup)
 STR - Strunzite (Group)
 SRT - Surite (Group)
 TAL - Talc (Family)
 trioctahedral (Supergroup)
 dioctahedral (Supergroup)
 TTA - Tetraauricupride (Supergroup)
 TTD - Tetradymite (Supergroup)

нм:c-1				
a=5.290Å				
b=9.173A c=9.460Å	o 🗕 🔒	9 0 0 0		
α=90.460°	• •			
β=98.680°	••• •			
γ=90.090°				
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Zeolite Classification

- Zeolite classification based on framework type
- These are noted by a three capital letter code describing network of corner sharing tetrahedrally coordinated framework atoms.

Zeolite Classification *	
BIK - Bikitaite	*
BOF - UCSB-15	
BOG - Boggsite	
BPH - Beryllophosphate-H	-
BRE - Brewsterite	=
BSV - UCSB-7	
CAN - Cancrinite	
CAS - Cesium Aluminosilicate (Araki)	
CDO - CDS-1	
CFI - CIT-5	
CGF - Cobalt-Gallium-Phosphate-5	
CGS - Cobalt-Gallium-Phosphate-6	
CHA - Chabazite	
CHI - Chiavennite	
CLO - Cloverite	
CON - CIT-1	Ŧ







Entries with Mineral Classification





Subfiles in Powder Diffraction File

Custom PDF Set Alkaloids Amino Acids, Peptides & Complexes Battery Material **Bioactivity** No Subclass Narcotic Psychotropic Carbohydrate Cement & Hydration Product Ceramic No Subclass Bioceramic Ferroelectric Microwave Perovskite Semiconductor **Common Phase** Education Explosive

Forensic Hydrogen Storage Materials

Inorganic Intercalate Ionic Conductors Merck Metals & Alloys Micro & Mesoporous No Subclass Clathrate Metal-Organic Framework Zeolite Mineral Related No Subclass Mineral Gem Natural Synthetic

Modulated Structure NBS Nucleosides & Nucleotides Organics Pharmaceutical No Subclass Excipient Pigment & Dye Polymer Porphyrins, Corrins & Complexes Steroids Superconducting Material No Subclass **Conventional Superconductor** Superconductor Reaction Product Superconductor Related **Materials** High Tc Superconductor Terpenes Thermoelectric Material



Subfile Growth by Release year











			HM : E a=7 b=12	P212121 .208Å 2.761Å	
🗟 C18 H21 N O5 -	05-006-3720		c=10 α=90 β=90	6.368A • 0.000° 0.000°	
File Edit Plots W	indow Help		γ=90	0.000°	
Save Print Preferen	nces Temperature Series Toolbox Property S	heet 2D 3D Bonds SAED EBSD Ring Simulated Profile Raw I	2013 Diffraction Data		
Cu Ka1 1.54056 Å	▼ Simulated Profile (Calc)	1,000			
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8.7794 10.0	063800 71 0 1 1 🔺				•
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PDF	Status: Primary QM: 0	Ovucadana N avida	∠ - OH 1,0	000	X-ray - Cu Kα1: 1.54056 Å (Simul.)
Experimental	Phase:	Oxycouolie /v-oxide		900	
Physical	Structural Formula: C18 H21 N O5			700	
, Crustel	Empirical Formula: C18 H21 N O5	Vijavakumar N. Sonar ^a Soan Parkin ^b and Poter A		500	
Crystal	Weight %: C65.24 H6.39 N4.23 (H ₃ CO O 4	100	
Optical	Atomic %: C40.00 H46.67 N2.22	Crooks**	(I) ³	800	
Structure	Compound Name: (5R,9R,13S,14S,17R)	1		
Classifications	Mineral Name:	"Department of Pharmaceutical Sciences, College of Pharmacy, University of	CH ₃		lipe Marchen and a strange a
	Common Name: Uxycodone N-oxide	Arkansas for Medical Sciences, Little Rock, AR 72205, USA, and "Department of	N		
Cross-references	CAS.	Chemistry, University of Kentucky, Lexington, KY 40506, USA	∠ { ОН	0 5 10 15 20 25	30 35 40 45 50 55 60 65 70 75 80 85 90
References	Entry Date: 09/01/2015	Correspondence e-mail: pacrooks@uams.edu			$ting 75 \mathcal{Y}_{20}$
Comments	Last Modifications:				Survey and the second s
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		Accepted 22 September 2012	H ₃ CO O OH	H ³ CO O	2016 - 1 Martin
		LINUNE IN LICONER 2017			INTERNATIONAL CENTRE FOR DIFFRACTION DATA

Modulated Structures

- Modulated structures form on account of perturbation in atomic position, occupation and/or thermal motion
- If period of fluctuation matches that of the 3D unit cell then a superstructure results otherwise an incommensurate modulated structure is obtained
- We can explain their symmetry in higher dimension 3+d (Super Space)



Schematic Representation





We can not index these satellite reflections in a regular 3D cell. It can be indexed by using a modulation vector q

$$\mathbf{Q} = h\mathbf{a}_1^* + k\mathbf{a}_2^* + l\mathbf{a}_3^* + m\mathbf{q}$$

$$\mathbf{q} = q_1 \mathbf{a}_1^* + q_2 \mathbf{a}_2^* + q_3 \mathbf{a}_3^*$$



Database Construct





Database Construct



Describes superspace. Can handle (3+1), (3+2) and (3+3) D cases



NaCO3 in average and modulated structure representation



Average Structure

Incommensurately Modulated Structure

$$q_1 = (0.182)a^* + (0.322)c^*$$



$\gamma NaCO_3$



Incommensurately Modulated Structure

$$q_1 = (0.182)a^* + (0.322)c^*$$



Comparison of Powder X-ray Diffraction Patterns



Neutron Diffraction Database

- As a part of PDF4+ database, first released in 2014 enabling search/match using CW neutron diffraction pattern
- More than 272,000 neutron diffraction entries in release 2016
- RIR based on neutron diffraction intensities for semi-quantitative phase analysis



Oilivine type LiFePO4 battery material



Raw Data Archival

- There are 11,284 raw powder diffraction data in release 2016 including crystalline, amorphous, nano materials
- This data can be exported out for analysis
- In future this feature will be extended to TOF and CW neutron patterns



Example of usage of raw data

TECHNICAL ARTICLE

The crystal structure of trandolapril, $C_{24}H_{34}N_2O_5$: an example of the utility of raw data deposition in the powder diffraction file

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¹Canadian Light Source, 44 Innovation Boulevard, Saskatoon, SK, S7N 2V3, Canada

²Illinois Institute of Technology, 3101 S. Dearborn St., Chicago, Illinois 60616

³Department of Chemistry, University College London, 20 Gordon Street, London WC1H 0AJ, UK

(Received 3 February 2016; accepted 13 April 2016)

The crystal structure of trandolapril has been solved by parallel tempering using the FOX software package with laboratory powder diffraction data submitted to and published in the Powder Diffraction File. Rietveld refinement was performed with the software package GSAS yielding orthorhombic lattice parameters of a = 19.7685(4), b = 15.0697(4), and c = 7.6704(2) Å (C₂₄H₃₄N₂O₅, Z = 4, space group $P2_12_12_1$). The Rietveld refinement results were compared with density functional theory (DFT) calculations performed with CRYSTAL14. While the structures are similar, discrepancies are observed in the configuration of the octahydroindole ring between the Rietveld and DFT structures, suggesting the refined and calculated molecules are diastereomers. © 2016 International Centre for Diffraction Data. [doi:10.1017/S0885715616000294]

Key words: trandolapril, powder diffraction, structure solution, density functional theory

Powder Diff. (2016) **31**(3),p 205-210





Debye-Menke Model: Valsartan ICDD: 00-0641634



Courtesy: Simon Bates, Triclinc Lab, Lafayette, IN: Presented at Denver X-ray Conference, Chicago August 1-5, 2016



On the Fly Simulations

TC316 % - 00-045-



Synchrotron



Conclusions

- In recent years there were several developments in Powder Diffraction File, including
 - Targeted subfiles
 - Enhancing classifications
 - Modulated structures
 - Neutron diffraction
 - Electron diffraction
 - Raw diffraction patterns
- Quality mark and editorial comments are useful while selecting a database entry
- Subfiles are very useful when performing search/match with a large database to minimize false positives



Thank you



